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July 2026

NAPIM Summer Course

Pigments for Ink Application

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Agenda

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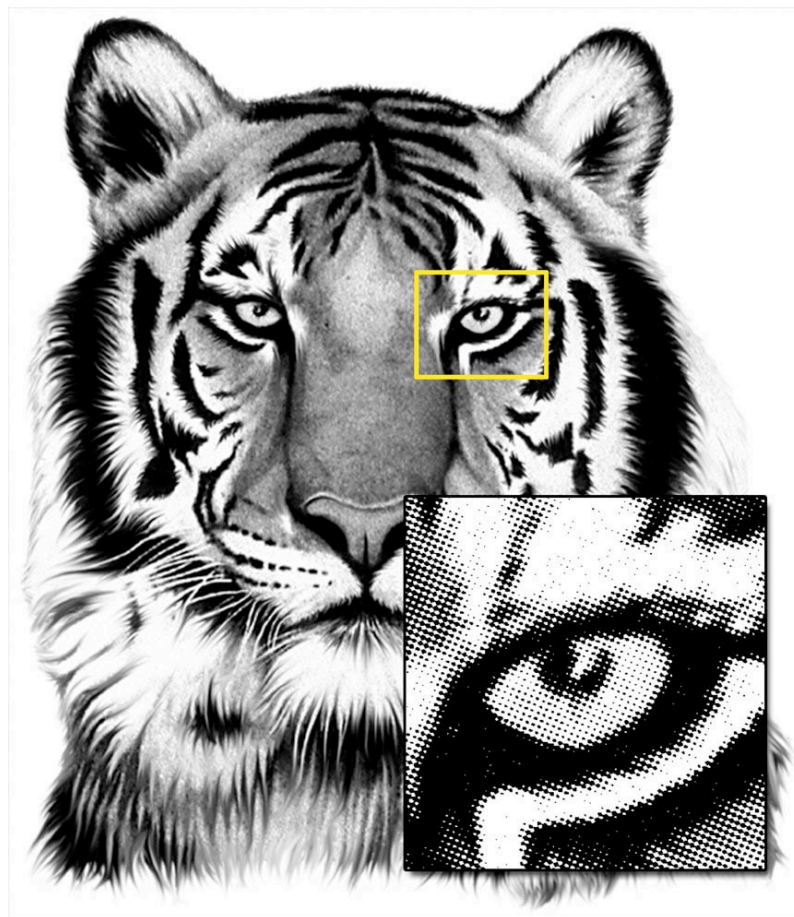
Line work

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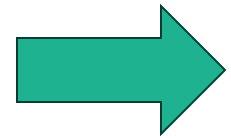
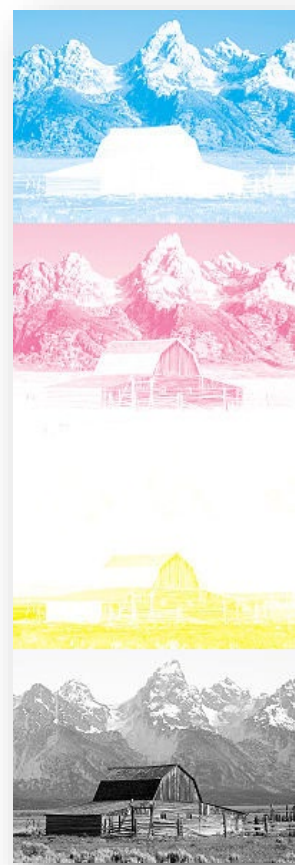


Image taken from www.greenlinepaper.com

Halftone dots



Process color



Some Common Printing Processes

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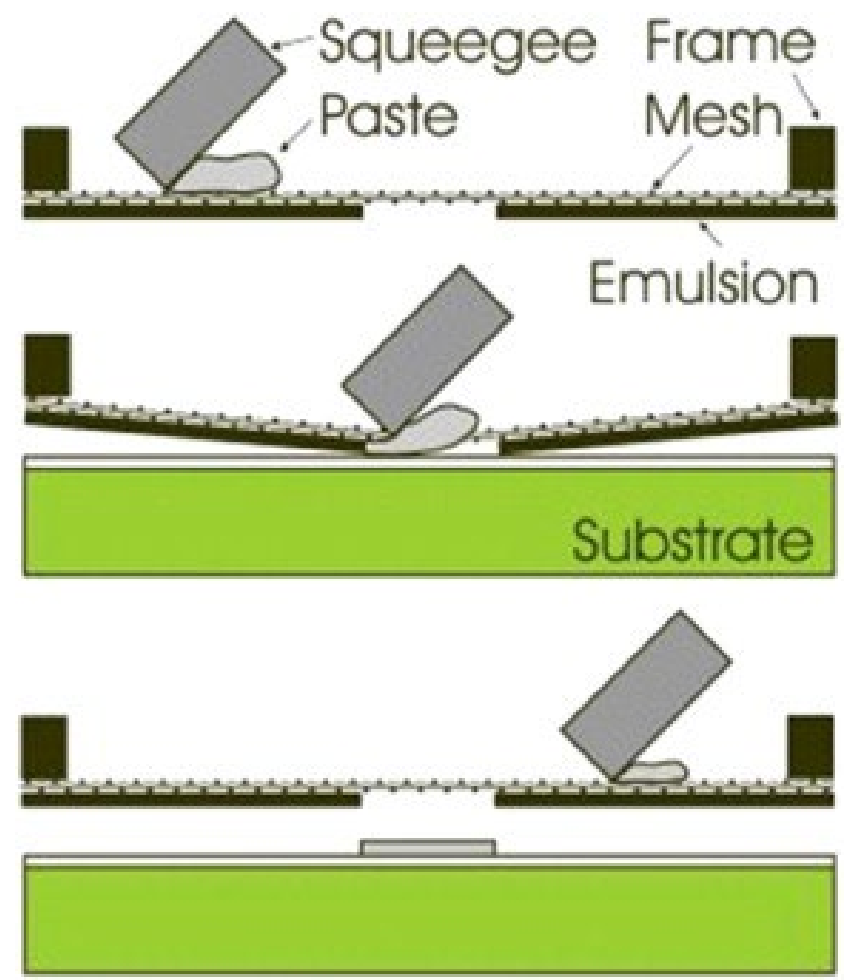
Screen Printing



Screen printing – process

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Screen printing – inks

- [illegible]

Lithography



Lithography – process



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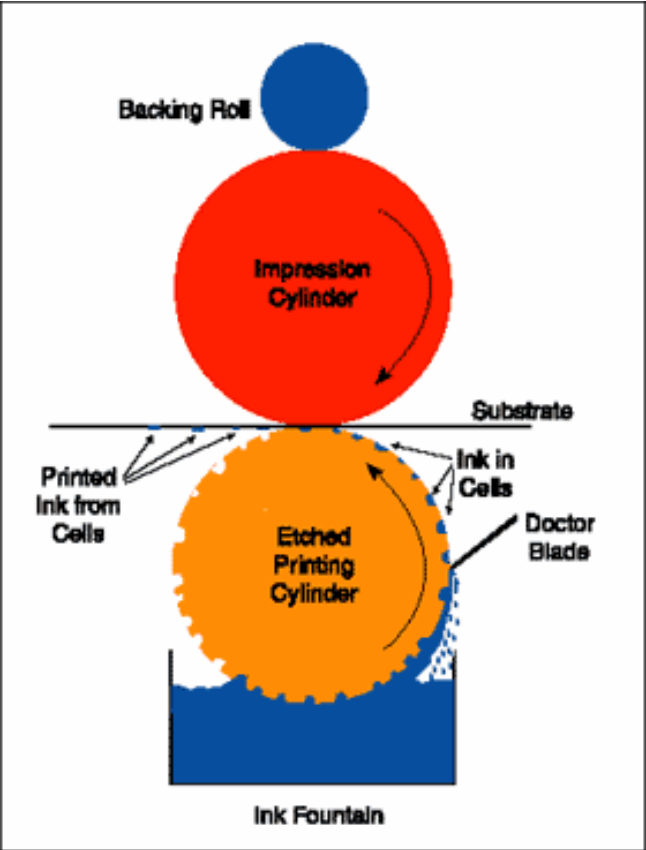
Lithography – ink properties

- [illegible]

Gravure

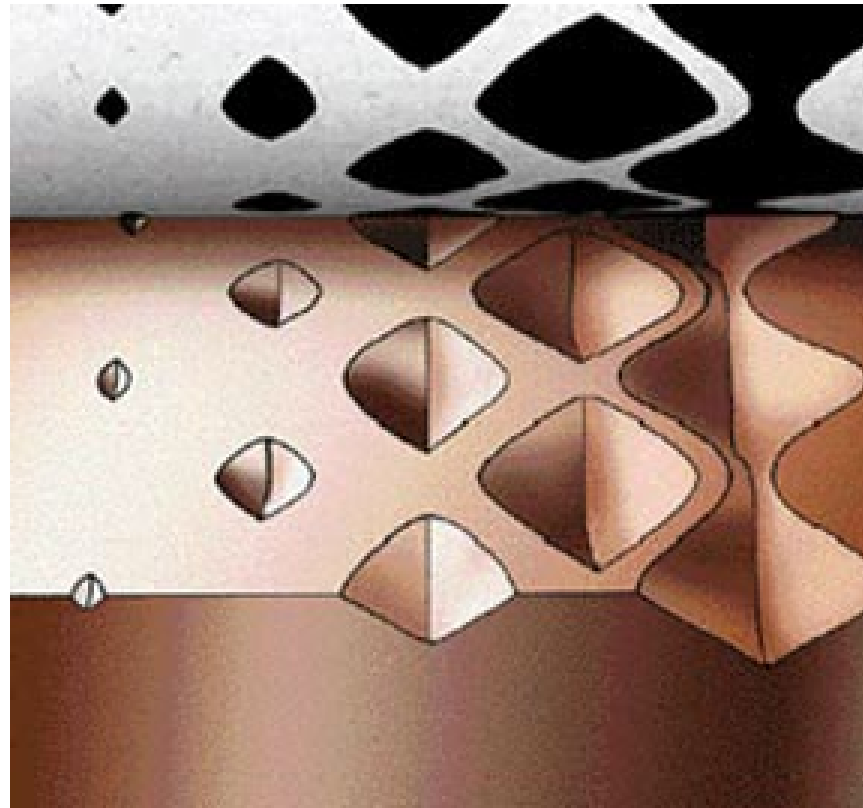


Gravure – process



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Gravure – engraved roll



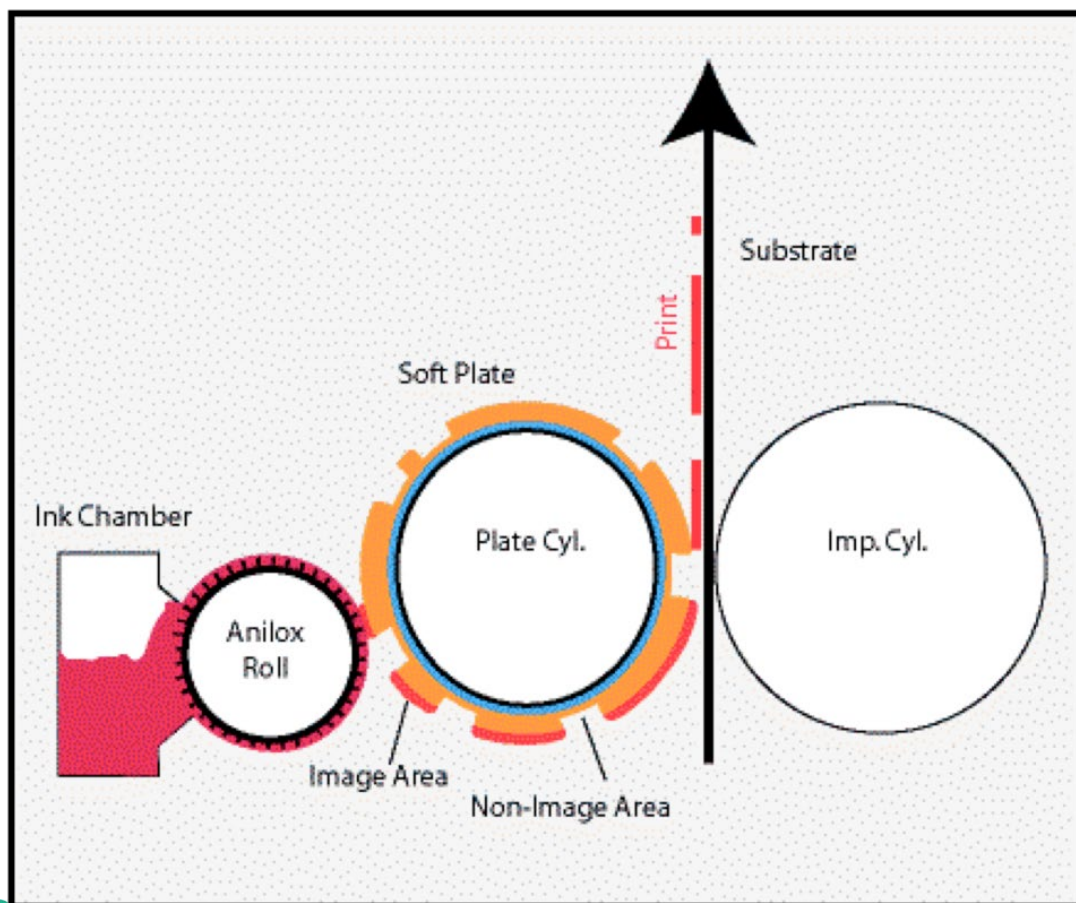
Gravure – ink properties

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Flexography



Flexography – process



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Flexography – ink properties

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Resolubility – why it matters



Dry Offset



Dry Offset

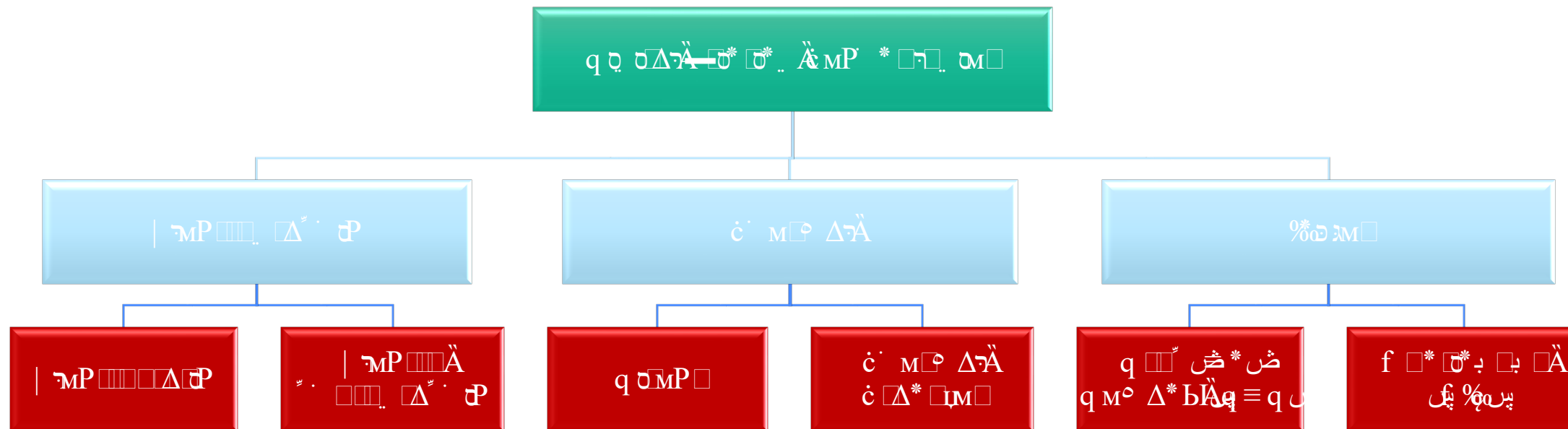
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Digital



Digital Printing Types: Overview



Drop-on-Demand Inkjet: Formulation Challenges

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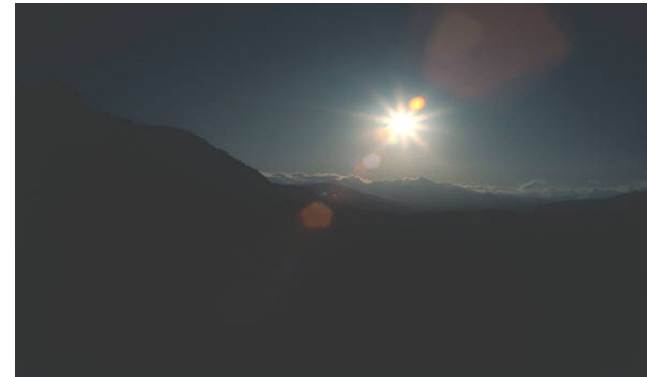
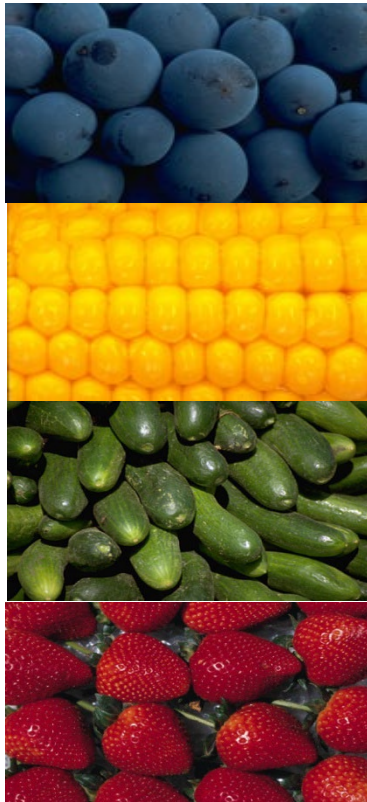
Printing Overview

	Gravure	Flexo	Lithography	Screen	Digital
Typical Print Speeds	Up to 600 m/min	~ 450 m/min	600-700 (heatset) 15000-18000 iph (SF)	100 m/min (rotary)	30-100 m/min (state of art)
Typical Ink Viscosities	20-25 secs Zahn 2 0.05 – 0.2 Pa.s	20-25 secs Zahn 2 0.05 – 0.5 Pa.s	40-100 Pa.s	1 – 5 Pa.s (flatbed) Lower for rotary	6-15 mPa.s solvent/UV 2-5 mPa.s aqueous
Typical Wet Film Thickness (μm)	7 – 15	3 – 7	0.5 – 2	8 – 30	6 – 12 (multiple passes)
Typical Particle Size	< 8 μm	< 4 μm	< 4 μm	-	Mean of 0.05 – 0.2 μm Max. of 0.5 – 1.0 μm
Main Binders	Nitrocellulose, Cellulose Acetate Propionate, Polyvinyl butyral, Polyamide, Polyurethane, Polyvinyl, Acrylics , Acrylated Oligomers	Nitrocellulose, Cellulose Acetate Propionate, Polyvinyl butyral, Polyamide, Polyurethane, polyvinyl, Acrylics , Acrylated Oligomers	Modified rosin esters (phenolic), Hydrocarbons, Drying oils (linseed oil, alkyds) Acylated Oligomers	Polyvinyls, Plastisol, Acrylated Oligomers	Polyvinyl, nitrocellulose, Acrylated Oligomers
Main Solvents	Alcohols (ethanol, n-propanol), Esters (ethyl acetate, n-propyl acetate), glycol retarders (ethoxy propanol), Water Ketones, Acrylated Monomers	Alcohols (ethanol, n-propanol), Esters (ethyl acetate, n-propyl acetate), glycol retarders (ethoxy propanol), Water Ketones, Acrylated Monomers	Petroleum distillates (heatset inks, 260-290°C), Mineral oils, Vegetable Oils (linsed, soya, rapeseed etc) Acrylated Monomers	Glycols, Acrylated Monomers	Water, Glycols, esters, ketones, Acrylated Monomers

What is Color?



What is Color?



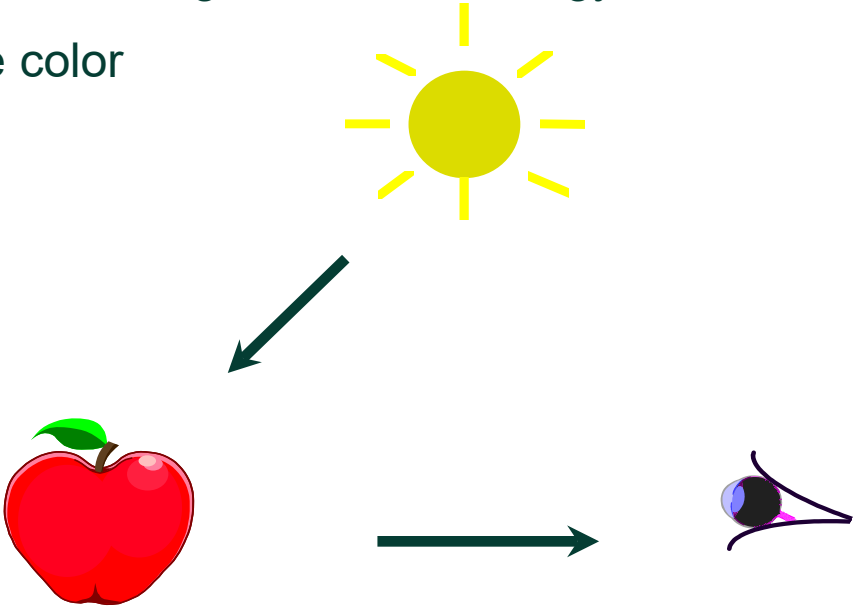
**Color is the physiological
perception of an object
exposed to light**

What does that mean?

No light —> no color

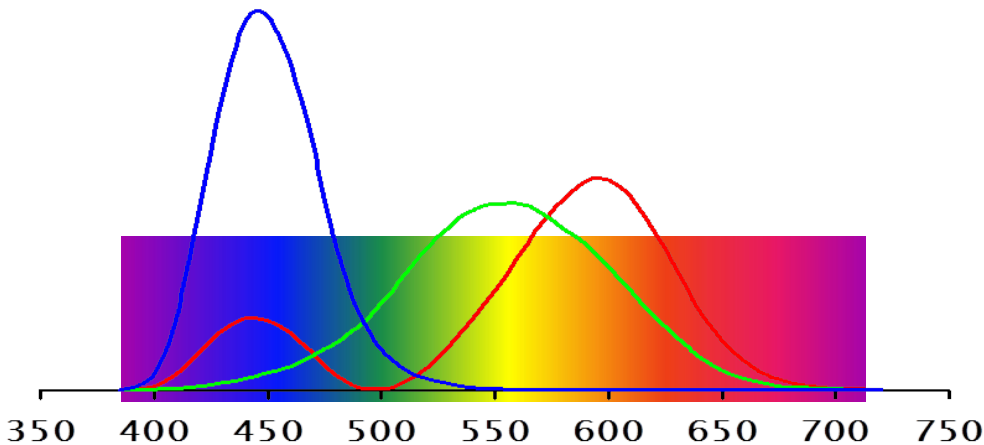
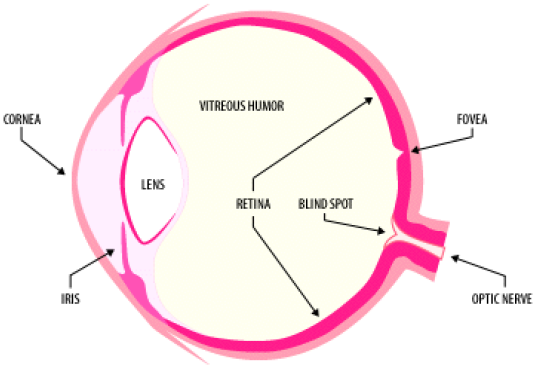
Three elements are necessary for perception of color:

- 1. **Light:** emitting energy
- 2. **Object:** absorbs a part of energy and re-emitting the rest as energy
- 3. **Receptor:** vision allows us to perceive color



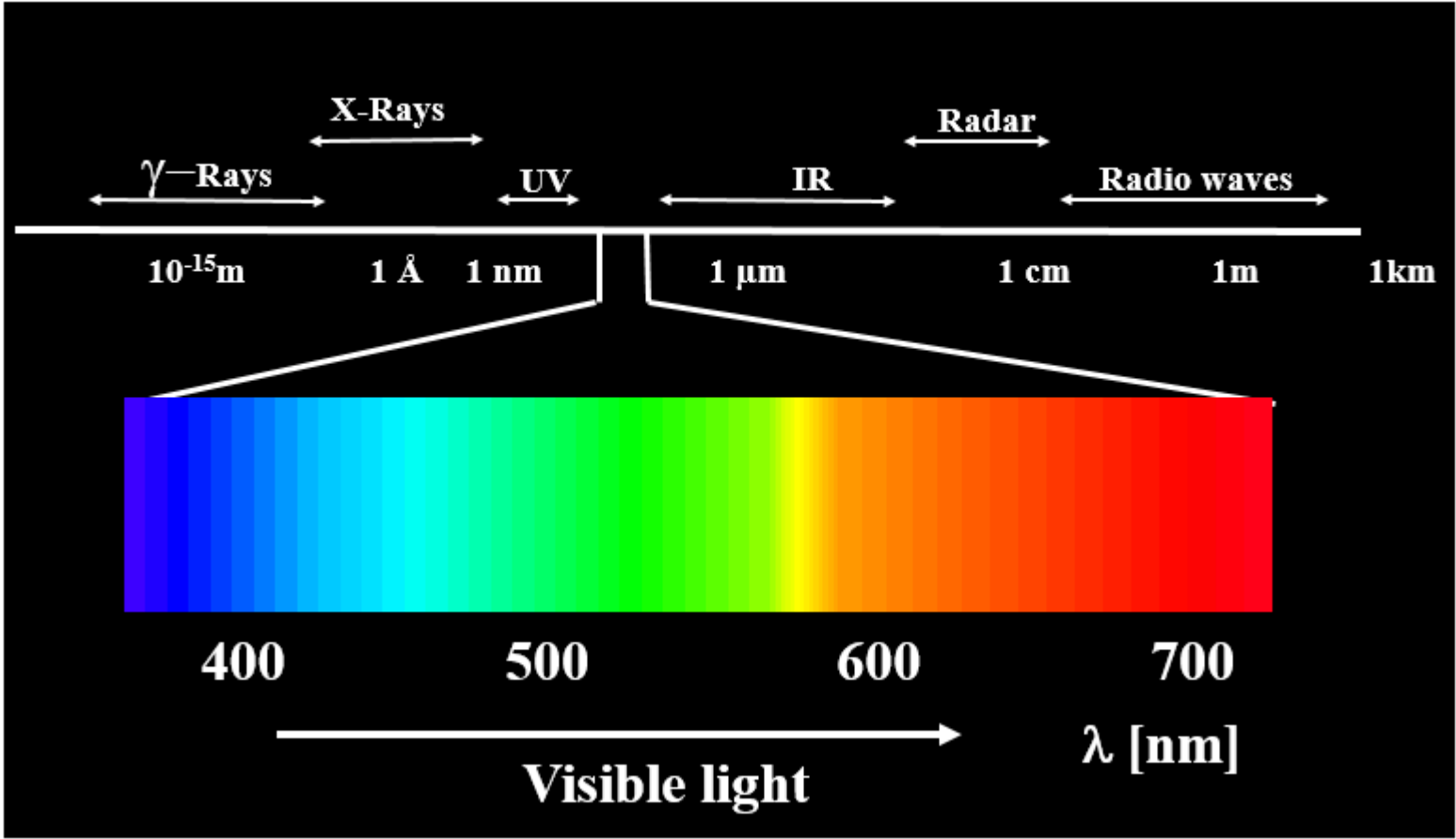
Perceiving Color – Color Vision

- Light enters the eye through the cornea, iris and lens and is imaged onto the retina.



- Light receptors, cones and rods, absorb some of the light and generate a signal that is interpreted by the brain.
- Cones are the receptors that are responsible for color.

Electromagnetic Spectrum

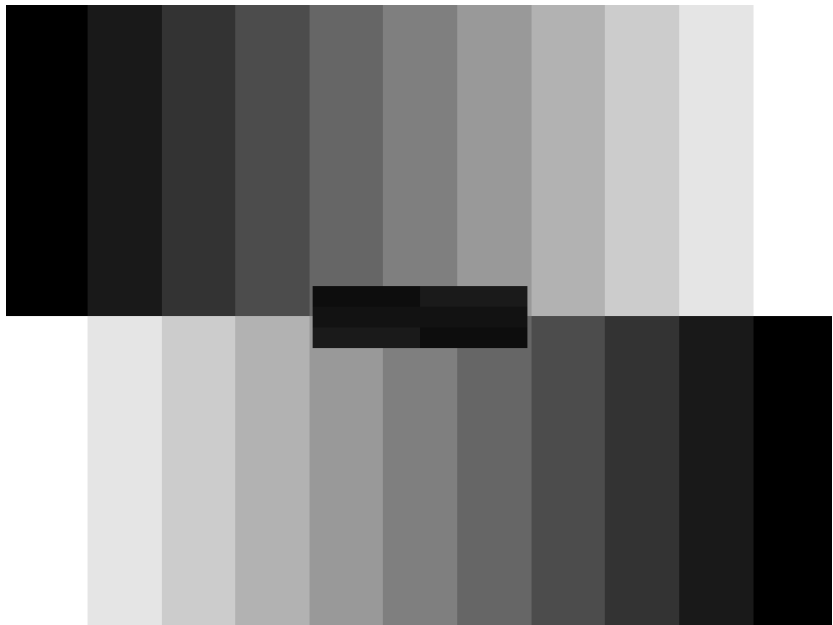


NRD/PG 2011

Describing Color

Lightness:

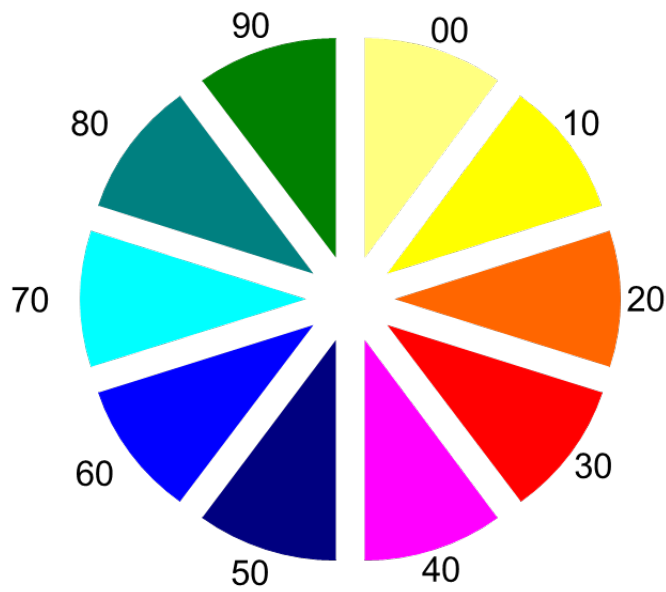
“Attribute by which a perceived color is judged to be equivalent to one of a series of grays ranging from black to white.” (ASTM E 284)



Describing Color

Hue:

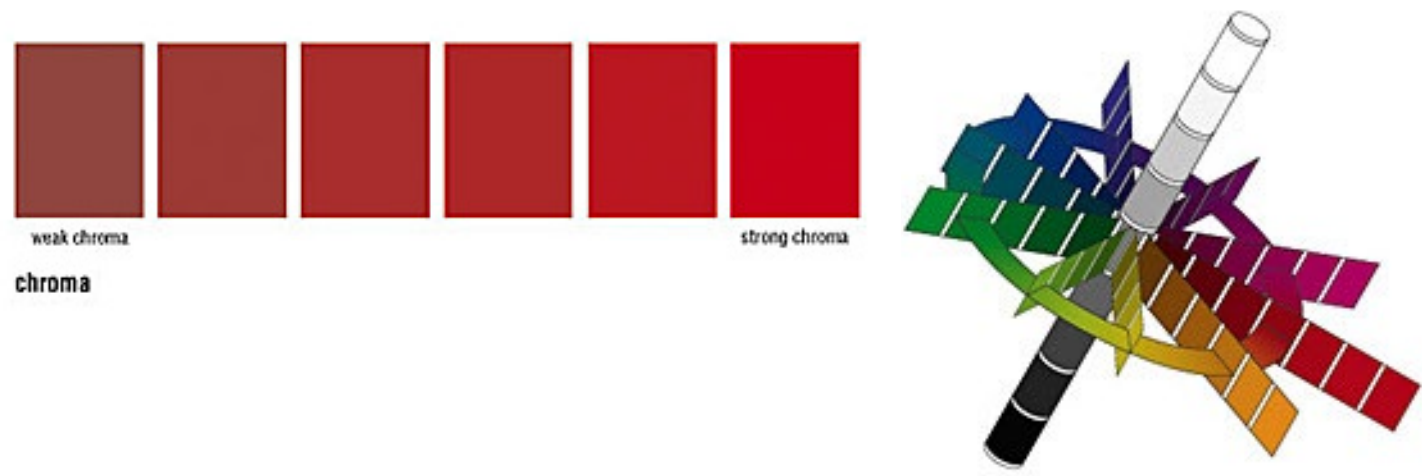
“Attribute of a visual perception according to which an area appears to be similar to one of the colors, red, yellow, green, and blue, or to a combination of adjacent pairs of these colors considered in a closed ring.” (CIE 17.4)



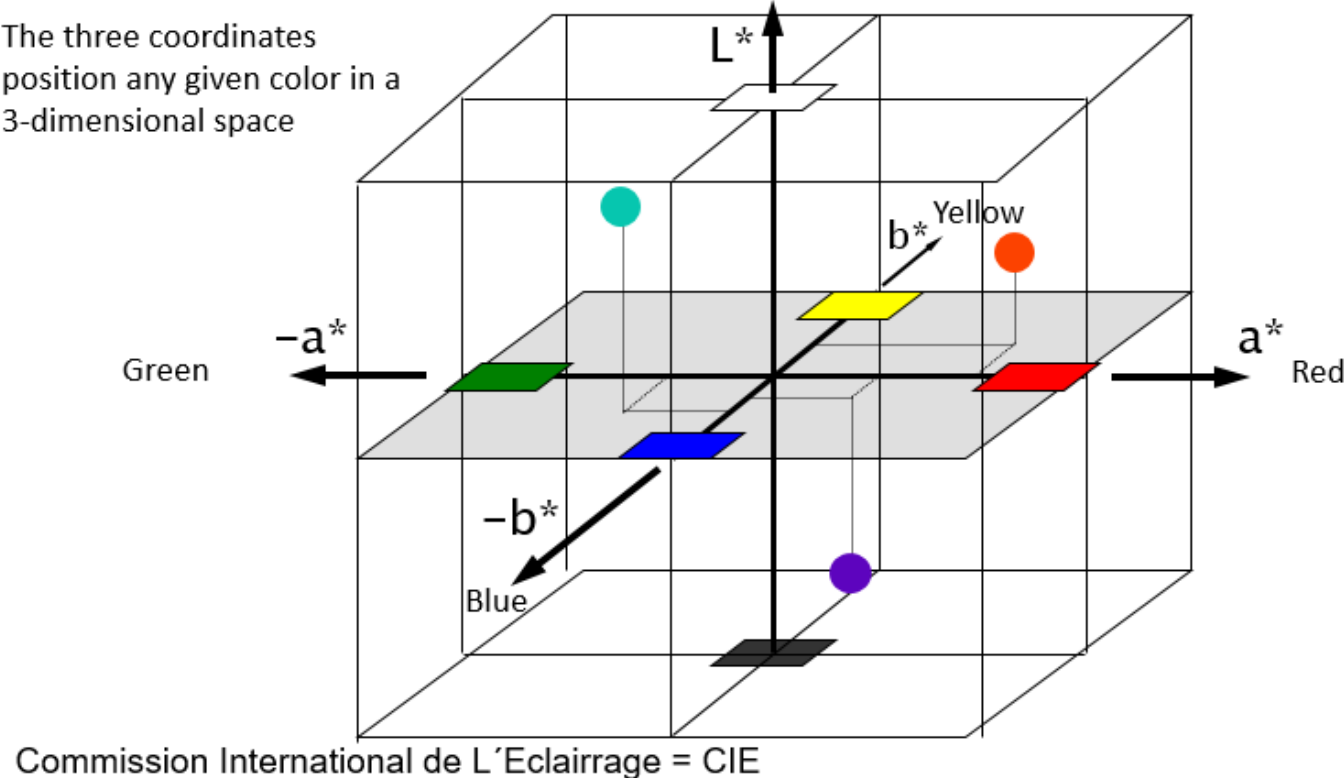
Describing Color

Chroma / Saturation:

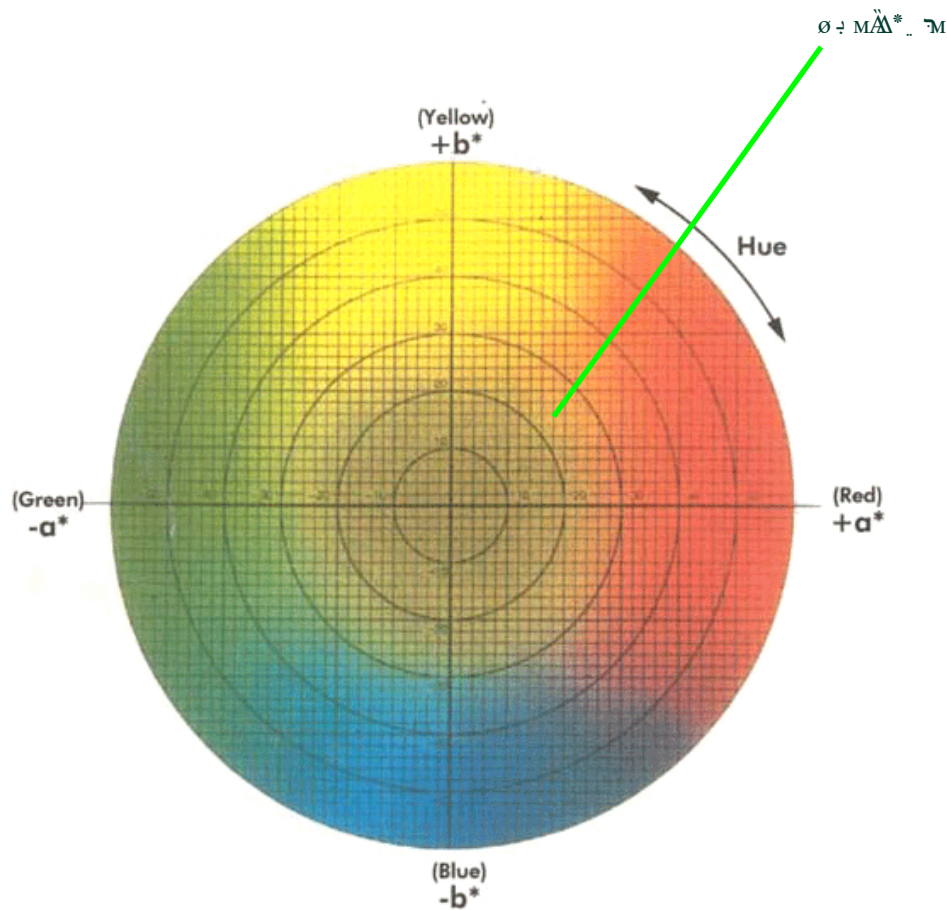
“Attribute of color used to indicate the degree of departure of the color from a gray of the same lightness.” (ASTM E 284)



(CIELAB) L*a*b* Color Space

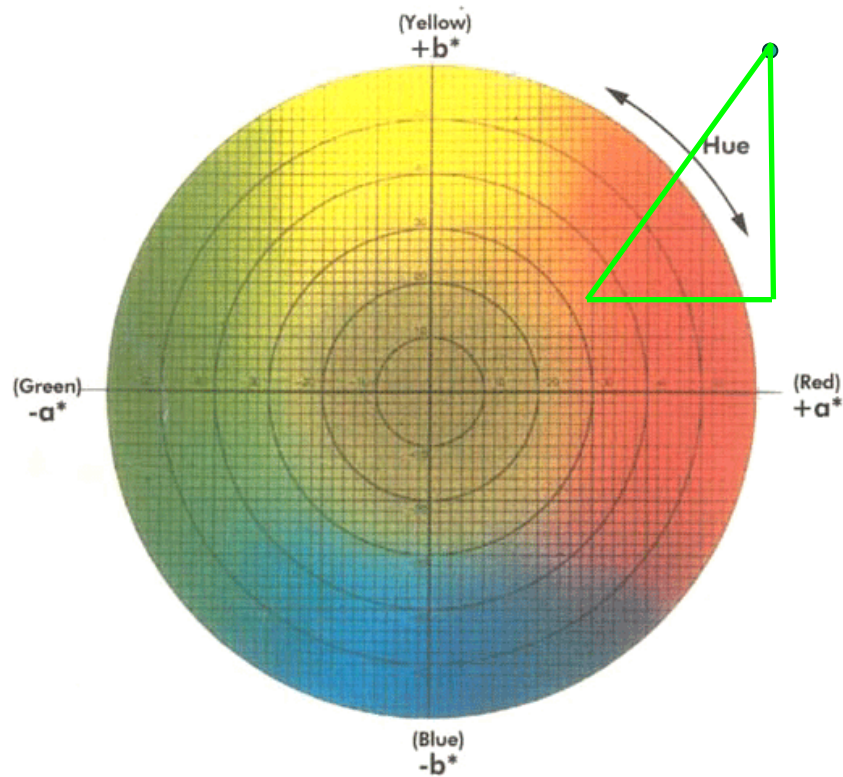


CIELAB Color Space



- Two additional coordinates, C^* and h , are calculated from L^* , a^* and b^*
- Hue, h , is a polar coordinate defining a color's position relative to the $+a^*$ axis (assigned 0°)
- Chroma, C^* , represents a color's distance from the neutral center
- C^* is the hypotenuse of the right triangle formed by a^* and b^*
- The hue angle can be determined through trigonometry

L*a*b* vs. L*C*h



- L^* , a^* & b^* and L^* , C^* & h are simply two ways of describing the position of the same point
- $L^* = 61.15$, $a^* = 30$, $b^* = 40$
- $L^* = 61.15$, $C^* = 50$, $h = 53.13^\circ$

- Most of the US tends to prefer using the $L^*a^*b^*$ coordinates

Pigment Properties

As a powder:

- Particle size
- Specific surface**
- Oil Absorption**
- Density
- Bulk density
- Dry content
- pH
- Conductivity
- Toxicity
- Safety properties
- Heat stability
- Inherent Compatibility**

In a medium:

- Hue
- Chroma
- ΔE^*
- ΔH^*
- Color strength
- Transparency
- Hiding power
- Flow properties
- Dispersibility
- Stability to flocculation
- Heat stability
- Fastness to overcoating
- Weather resistance
- Gloss

Opacity/Transparency

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What is Metamerism?

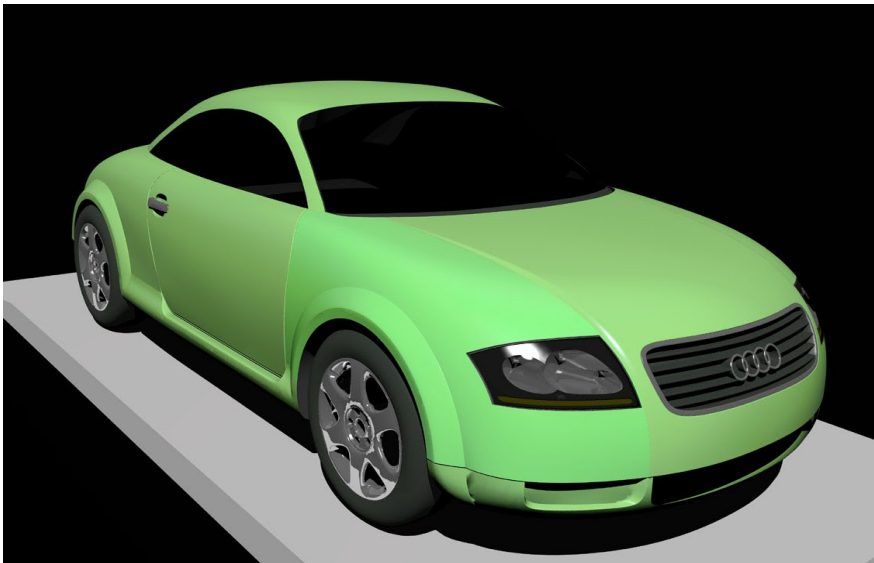
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Metamerism is the phenomena in which two objects seem to have the same color under one light source, but different colors under another light source

Metamerism is caused by using different pigment chemistry or materials in the paint formulation



Illuminant F11 (TL84)

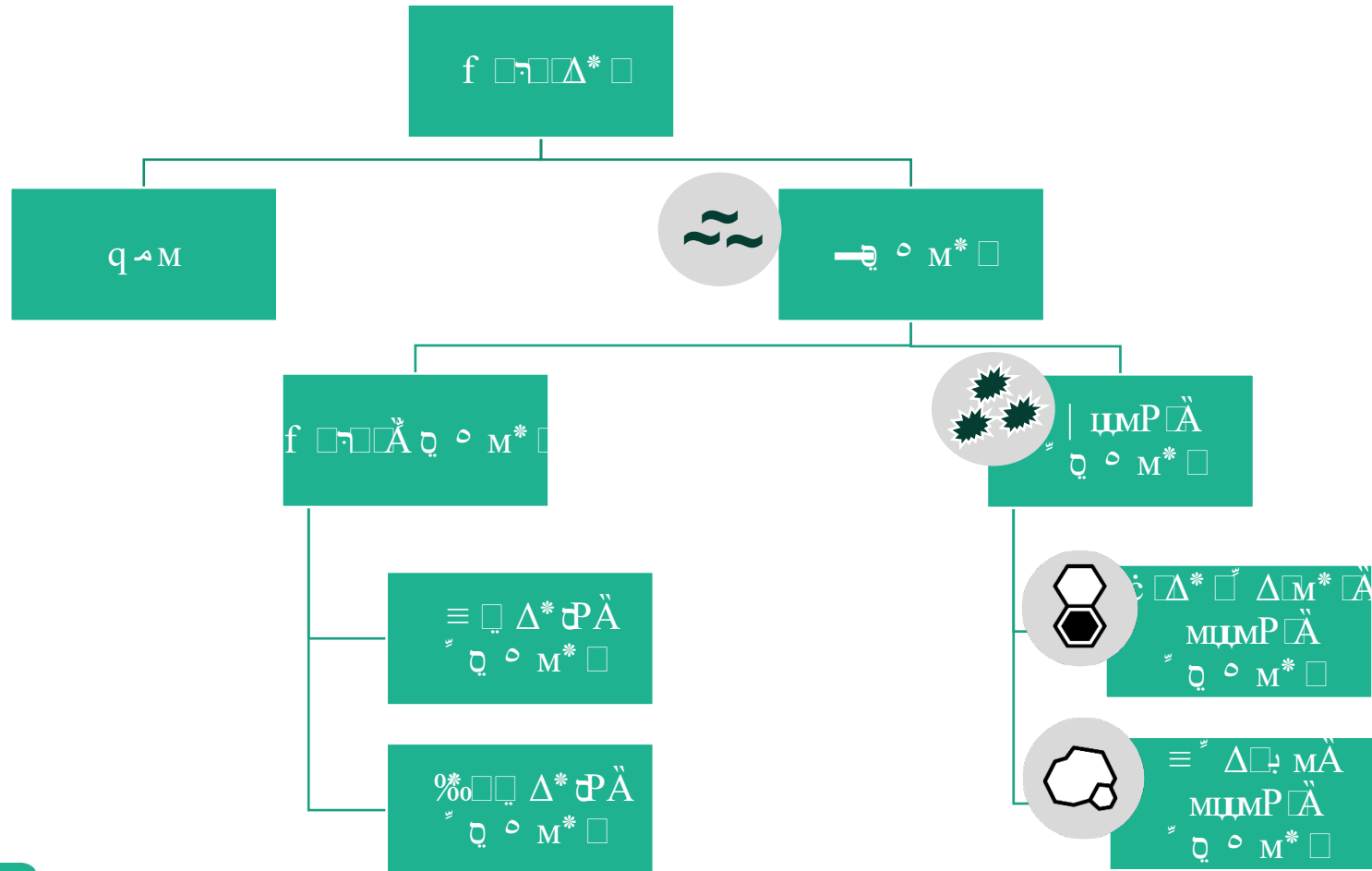


Illuminant D65

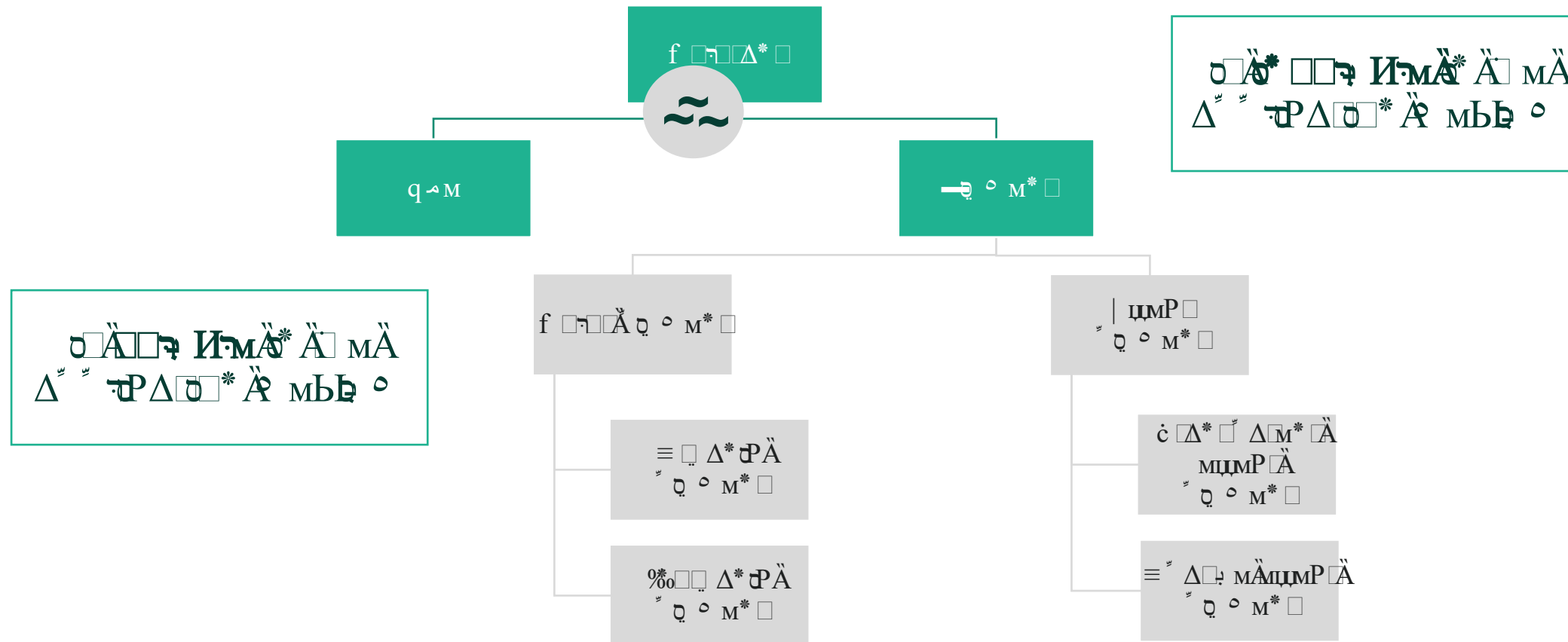
Colorants for Inks



Types of colorants

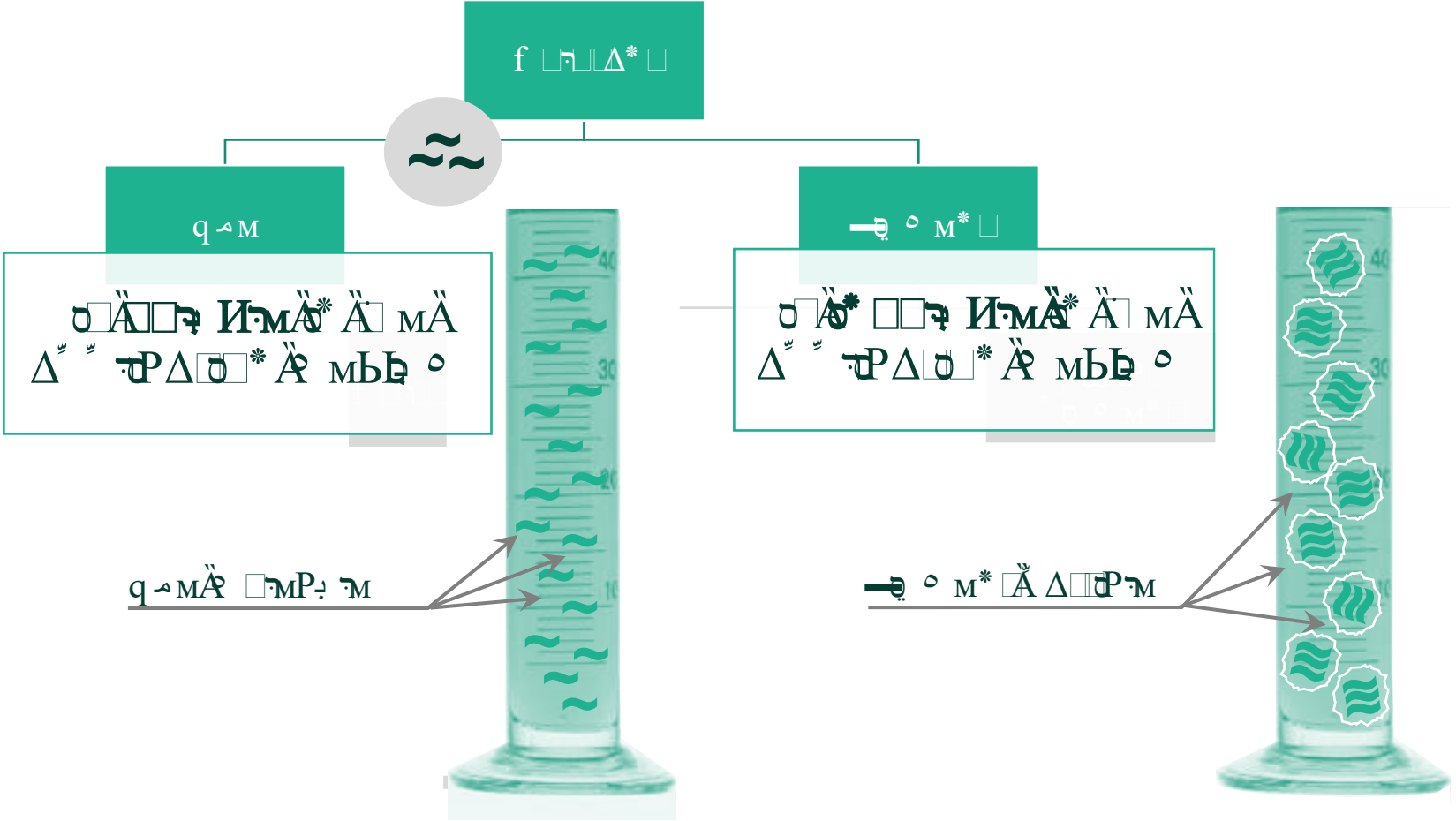


What is a dye? What is a pigment?



What is a dye?

What is a pigment?



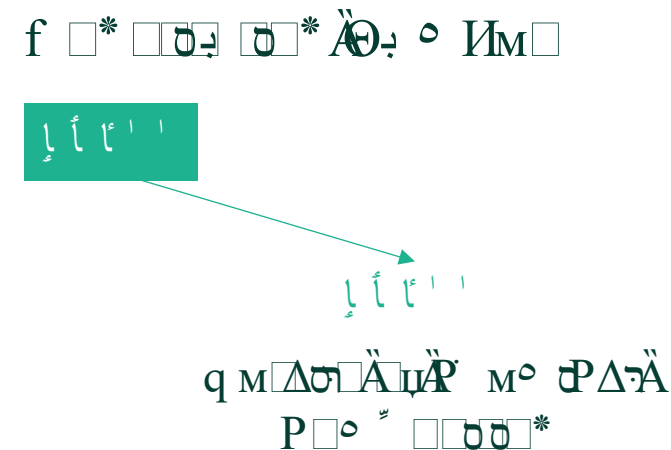
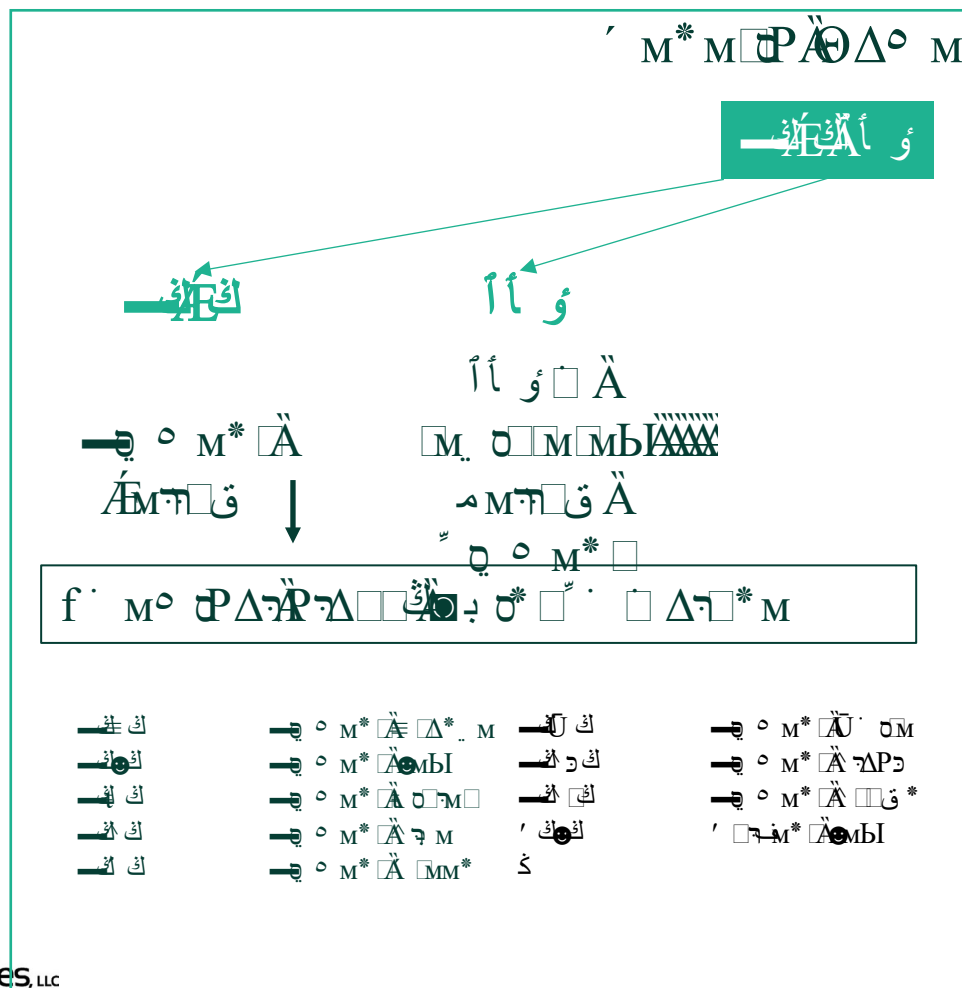
What is a dye?

What is a pigment?



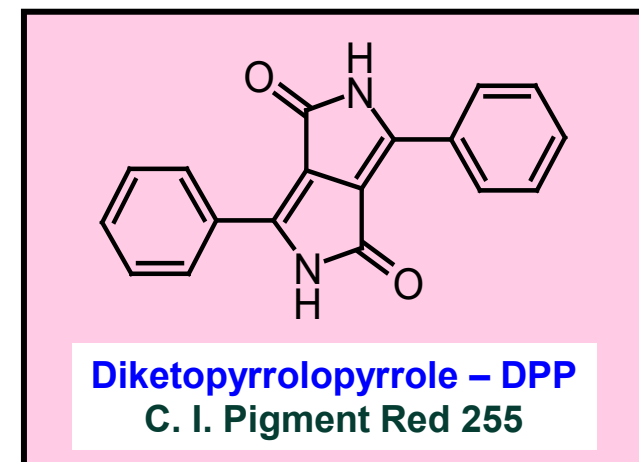
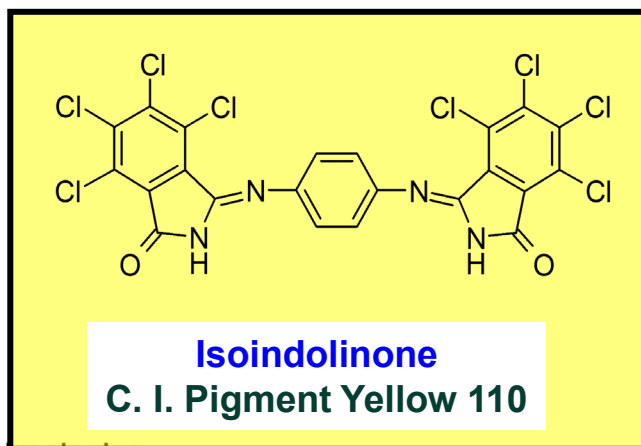
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What is Color Index?



Color Index (CI)

- [illegible]



C.I.Pigment Violet 19



**Cinquasia®
Red
L 4100 HD**

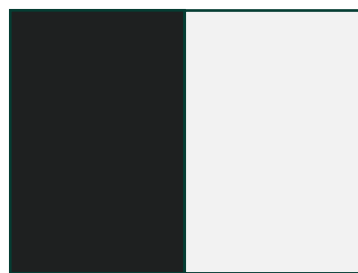
**Cinquasia
Violet
L 5120**

Mass Tone

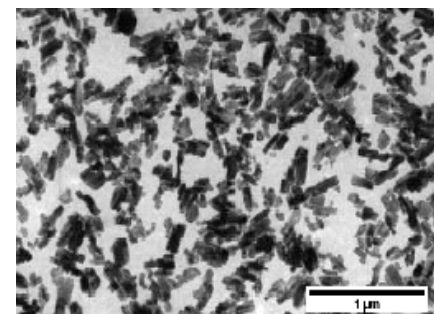
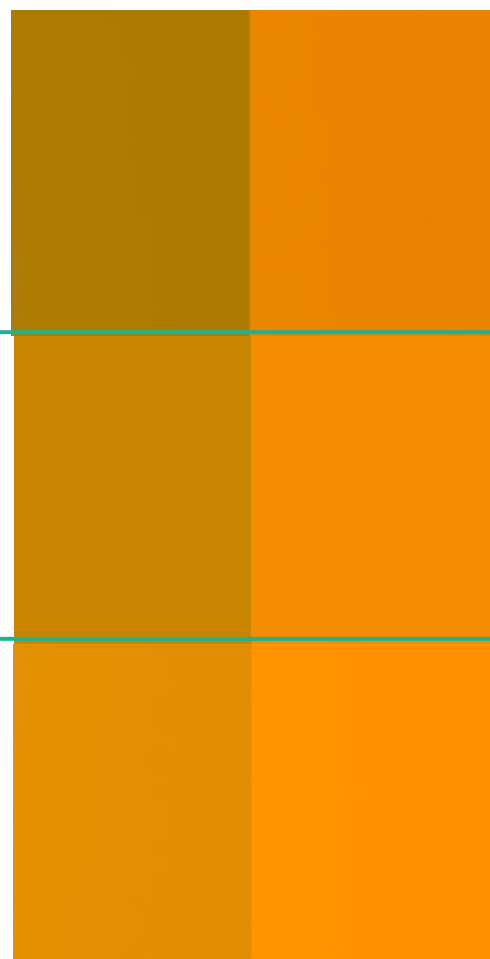
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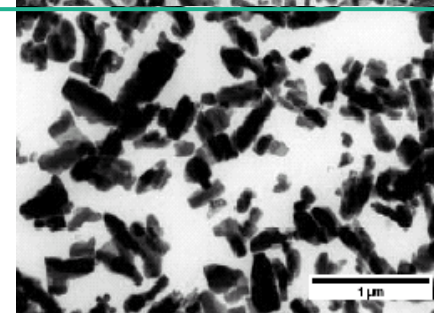
PY 139 Examples



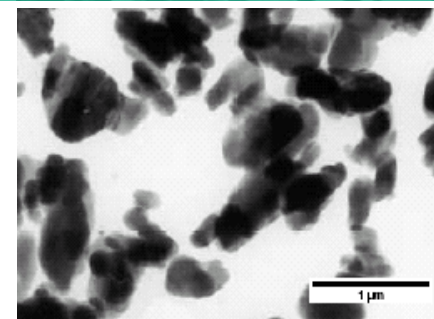
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~130 nm

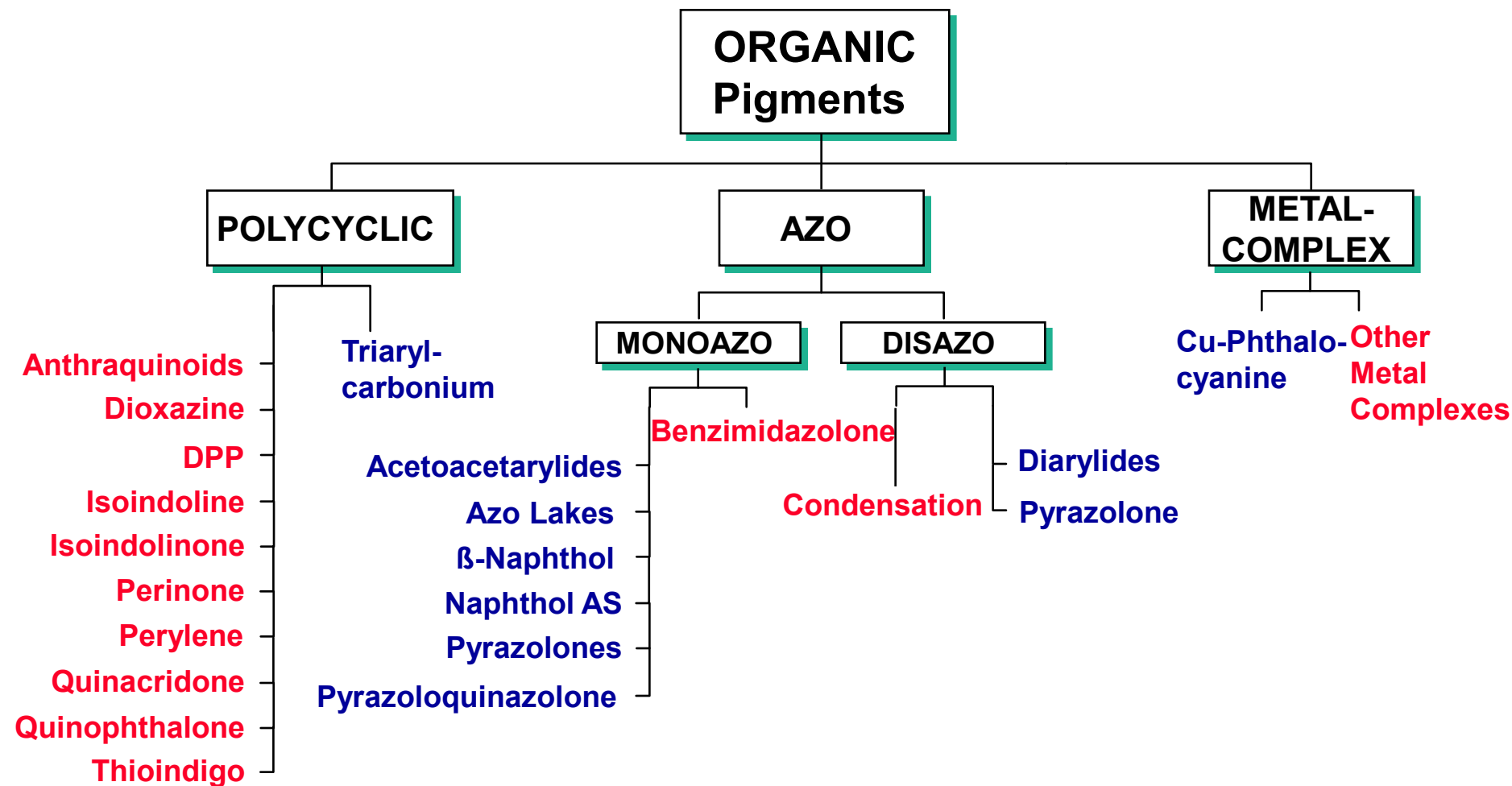


~200 nm

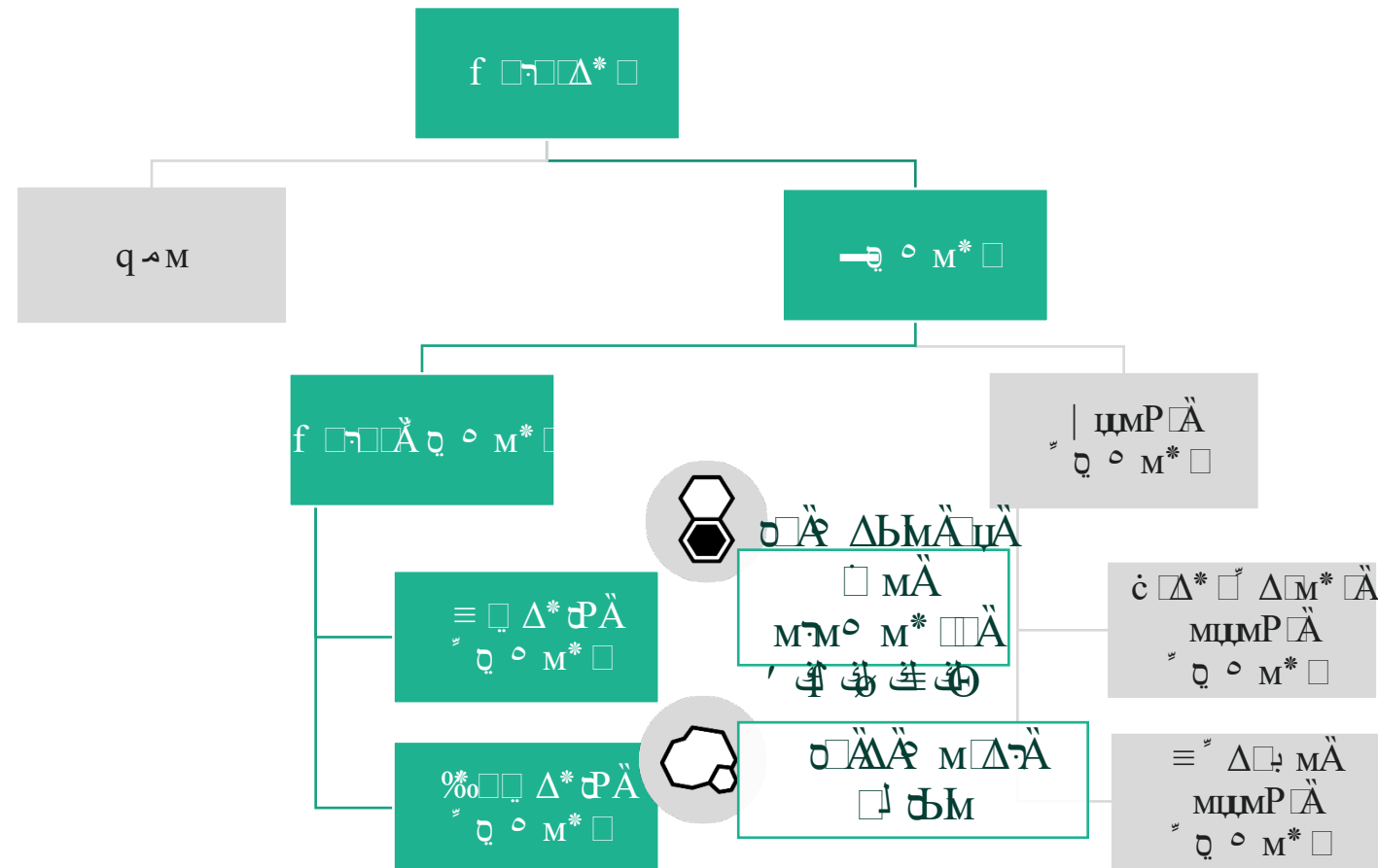


~300 nm

Organic Pigments

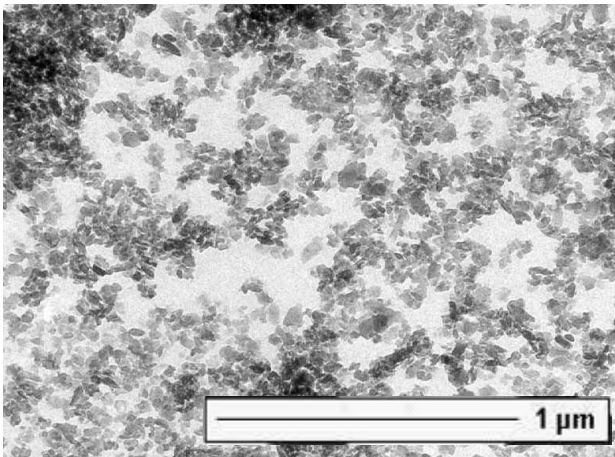


What is the difference between organic and inorganic pigments?

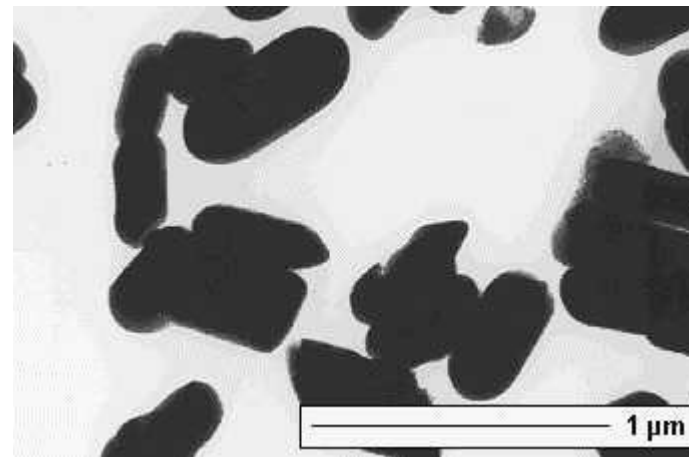


What do they look like under a microscope?

Micrograph of a material with a granular texture.

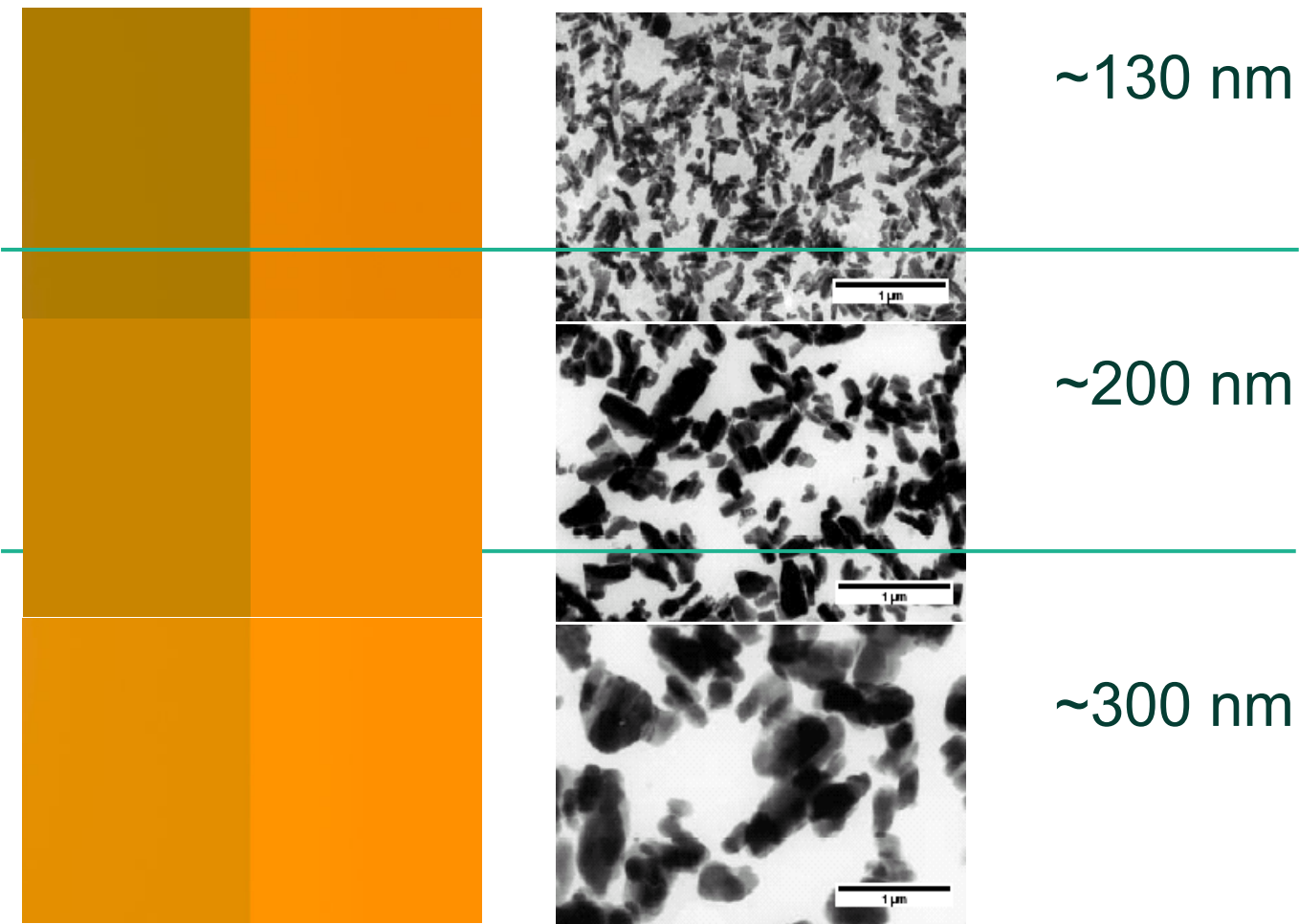


Micrograph of a material with a granular texture.



Micrograph of a material with a granular texture.

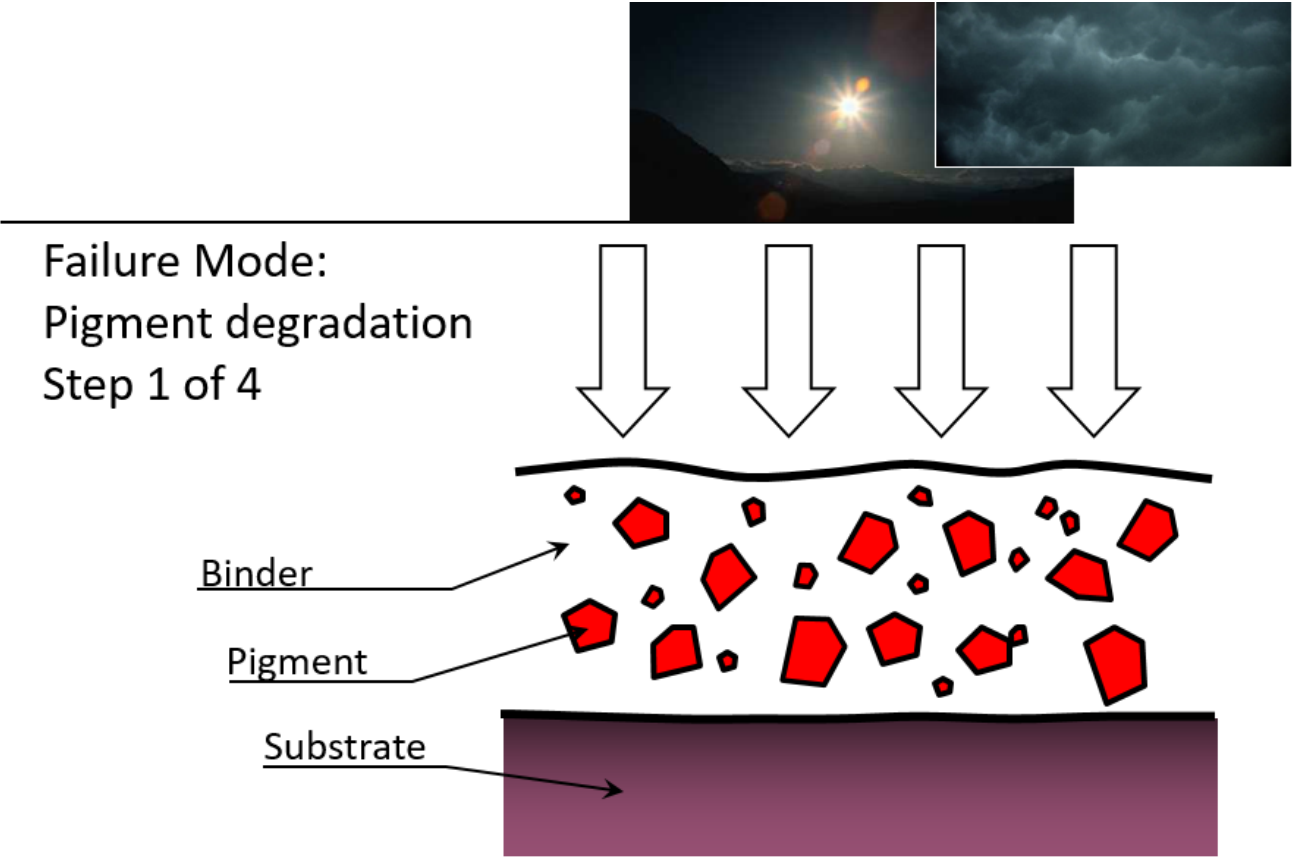
Influence of particle size on transparency and opacity



How do properties of organic and inorganic pigments relate to particle size?

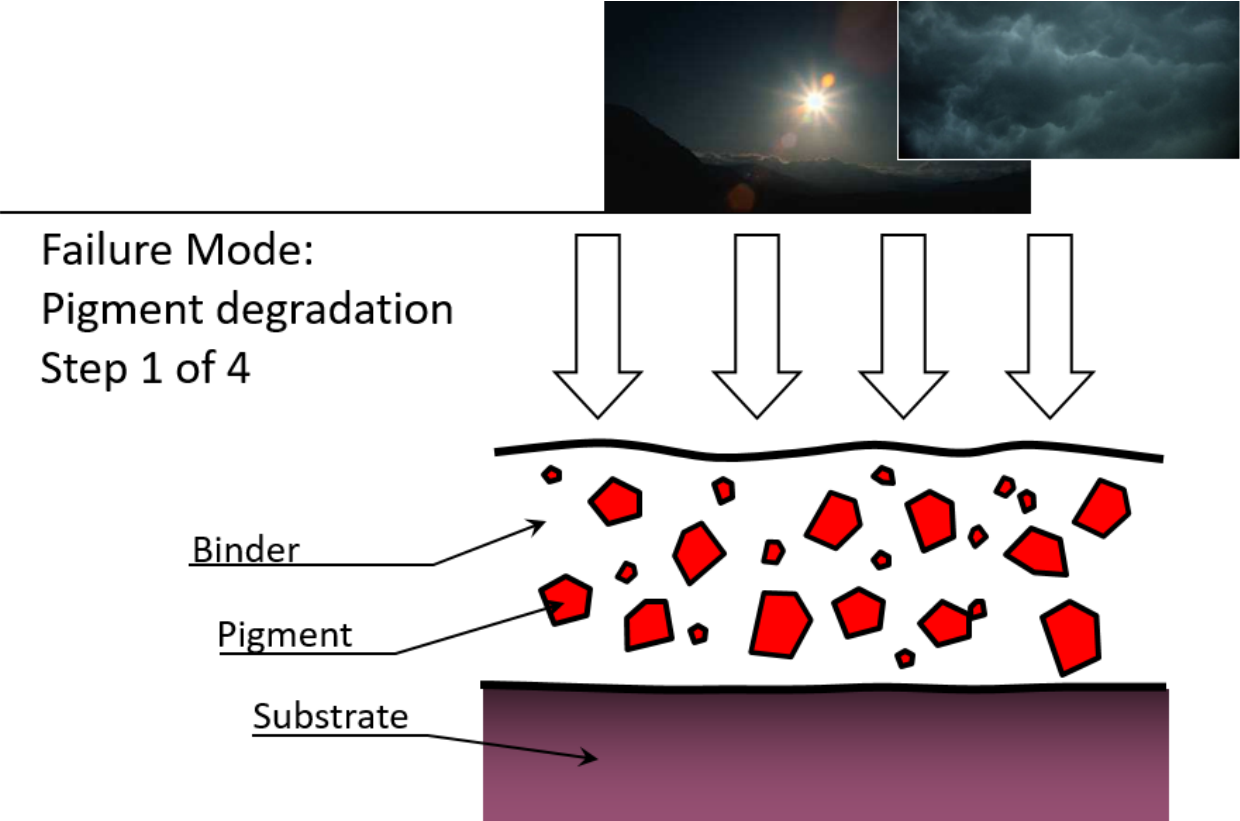
Property	Organic 	Inorganic 
Particle size		
Transparency		
Opacity		
Color strength		
Chroma		

Fastness Properties



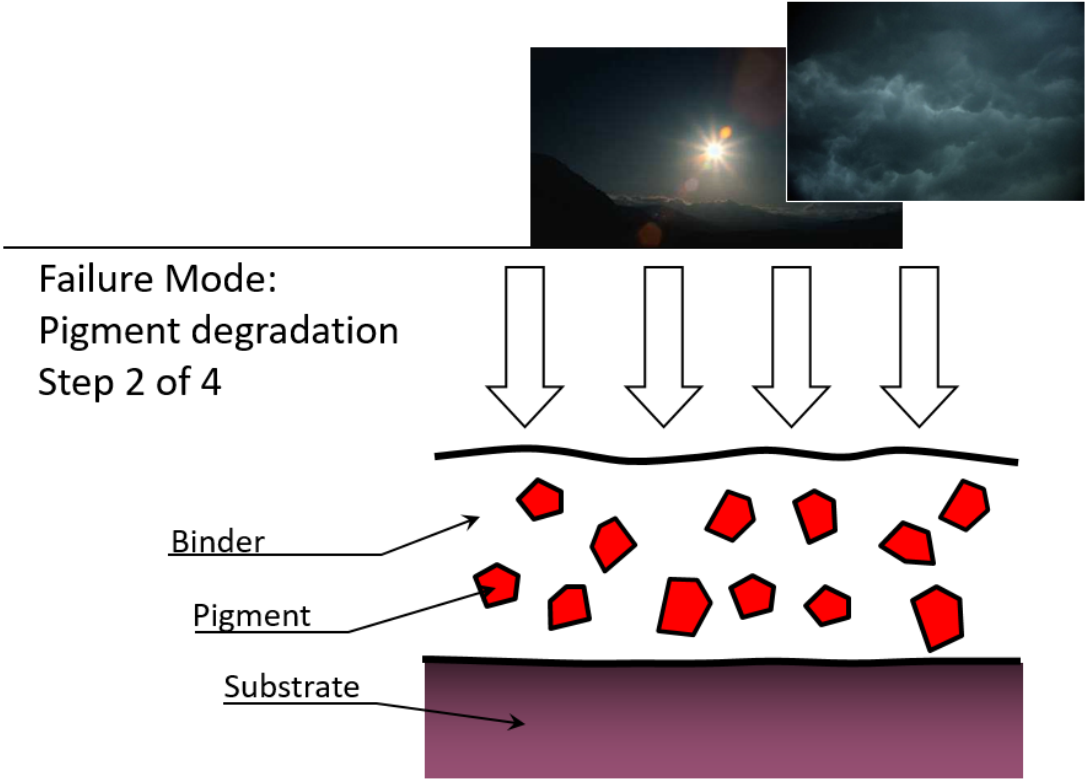
Fastness properties

e.g. weatherfastness



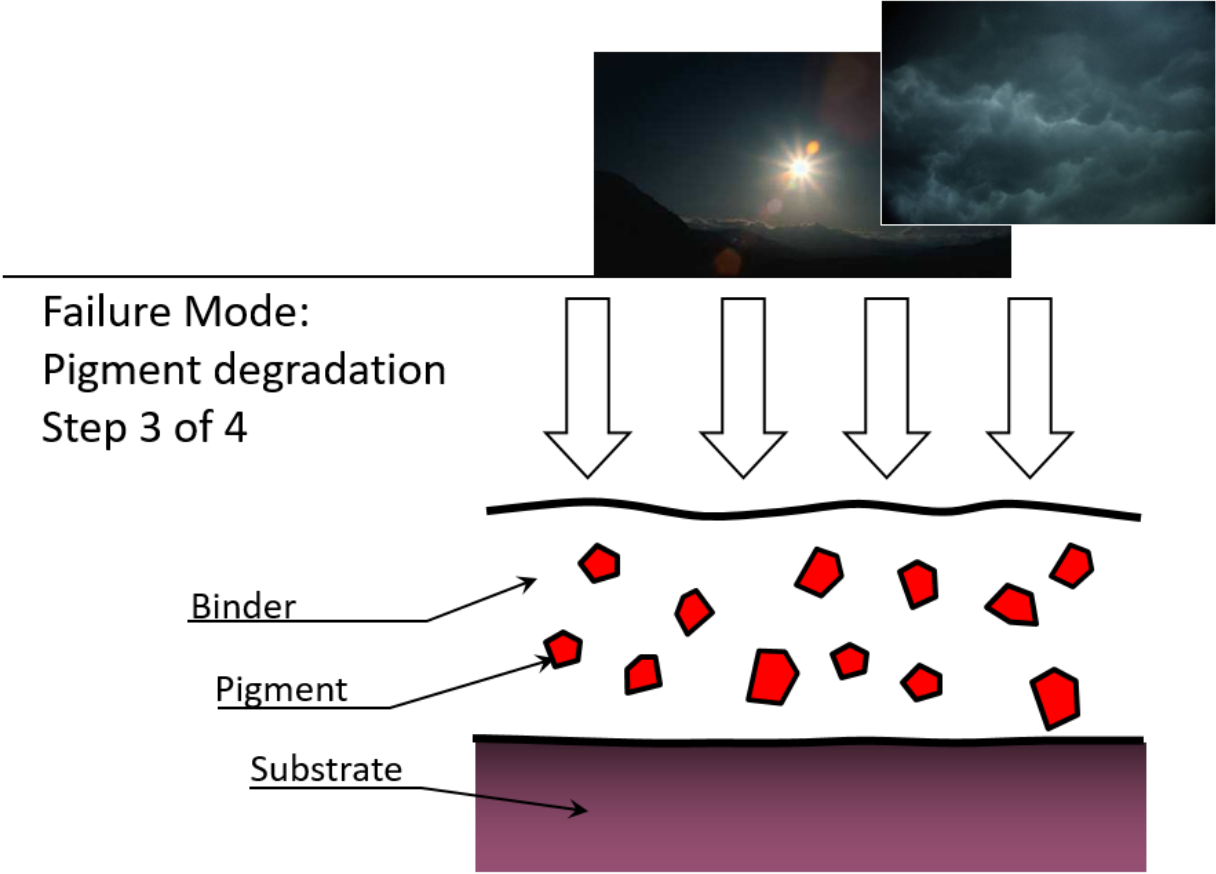
Fastness properties

e.g. weatherfastness



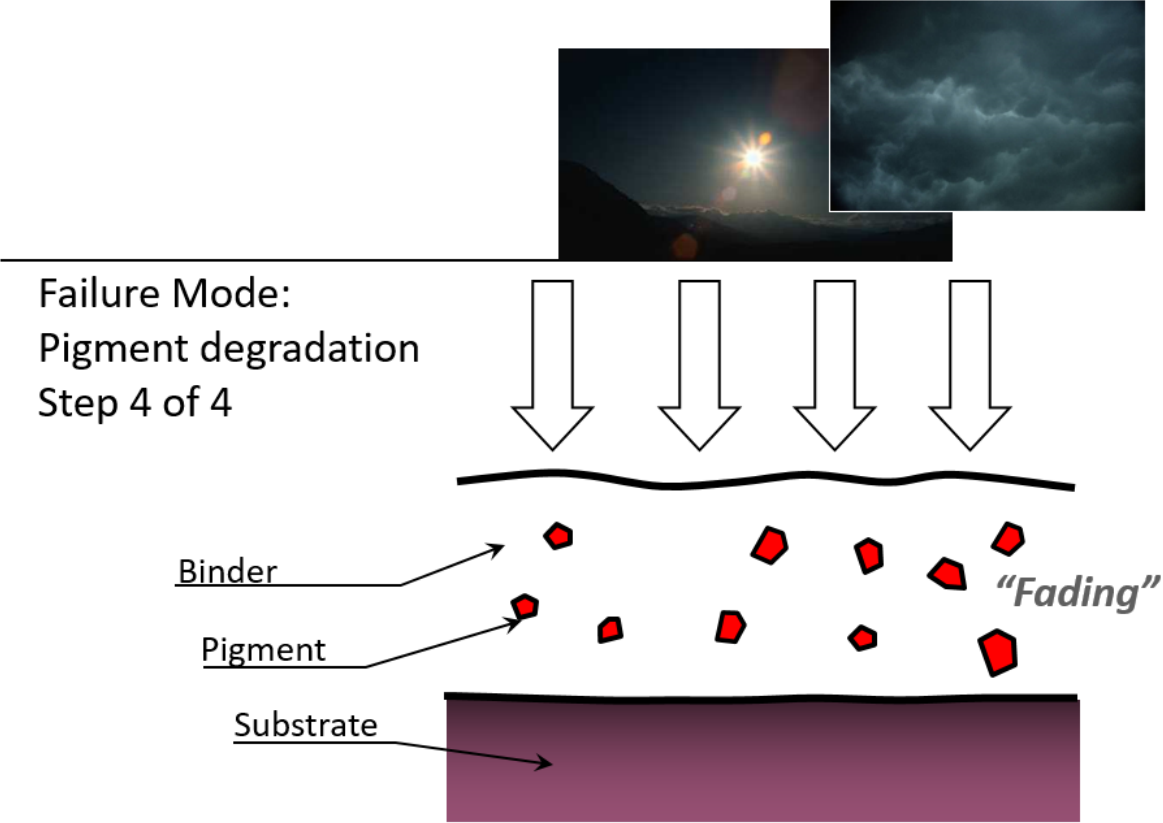
Fastness properties

e.g. weatherfastness



Fastness properties

e.g. weatherfastness



How do properties of organic and inorganic pigments relate to particle size?

Property	Organic 	Inorganic 
Particle size		
Transparency		
Opacity		
Color strength		
Chroma		
Fastness properties		

Pigments and Performance - Overview

Pigment Properties	Inorganic	Organic Classical	Organic High Performance
<u>Color</u> , Purity	Often Dull	Usually Bright	+/- High
Opacity	High	+/- transparent	+/-transparent
<u>Color</u> Strength	Middle to Low	Normally High	Usually Bright
Lightfastness (Blue Wool Scale)	Good to High (7-8)	Low to Middle (<7)	Good to High (7-8)
Weather Resistance	Varies/Chemistry	Insufficient	Middle to High
Heat Resistance	> 500°C	150°C to 220°C	200°C to 300°C
Solvent Fastness	High	Middle to Good	Good to High
Chemical Resistance	Varies/Chemistry	High except salts	High
Product cost	Low to Middle	Middle	High

Pigment Manufacture



' M* M□Δ.Π

☉Δ ق Ä ΔM□Δ.□

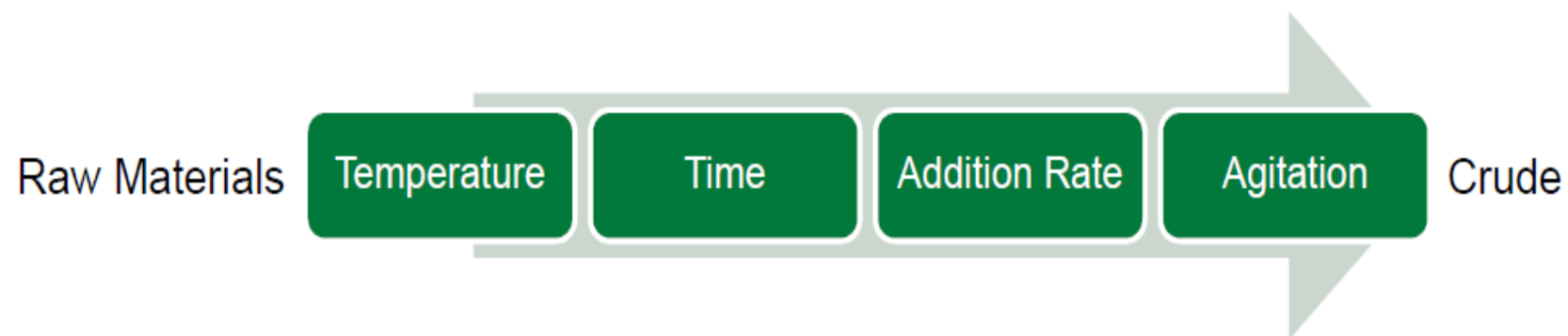
f· m° ρΔ.Ä~* □ m□□

f □* ЁБ□□*σ*.. ÄÄ~ m□ρΔ.η

f □* ЁБ□□*σ*.. ÄÄ· m° ρΔ.η

βσ*□ □σ*..

Synthesis



Chemical and Physical Conditioning



Finishing

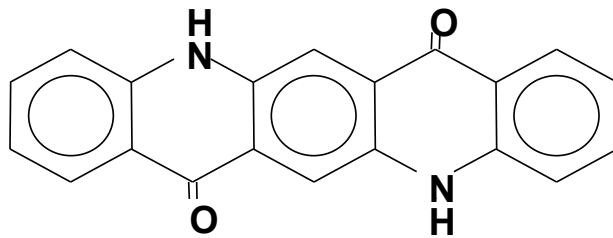


Organic Pigments - Auxochromes

- [illegible]



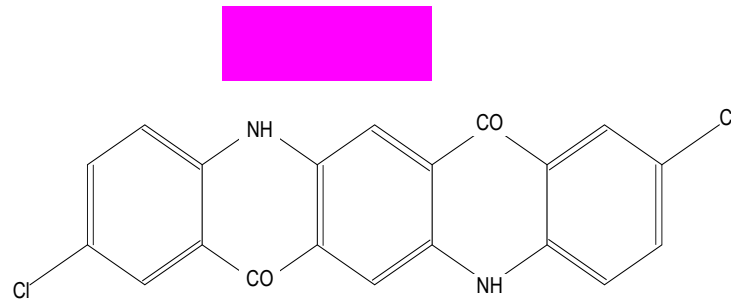
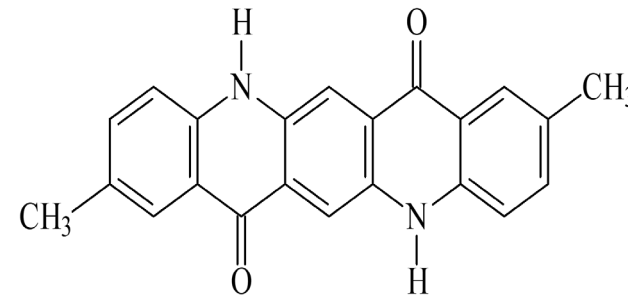
Organic Pigments: Auxochromes



Pigment Violet 19



Pigment Red 122

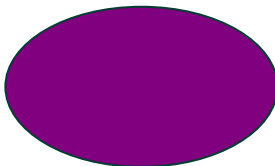
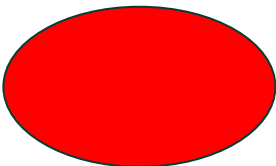


Pigment Red 202

Pigment Crystallography

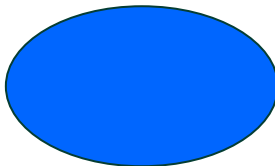
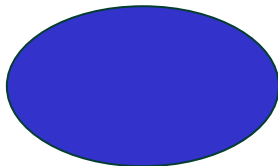
Polymorphism

Arrangement of a molecule can be different in the crystal lattice. This leads to different pigments with different colour and performance :



β form – reddish violet

γ form – bluish violet



α form – reddish blue

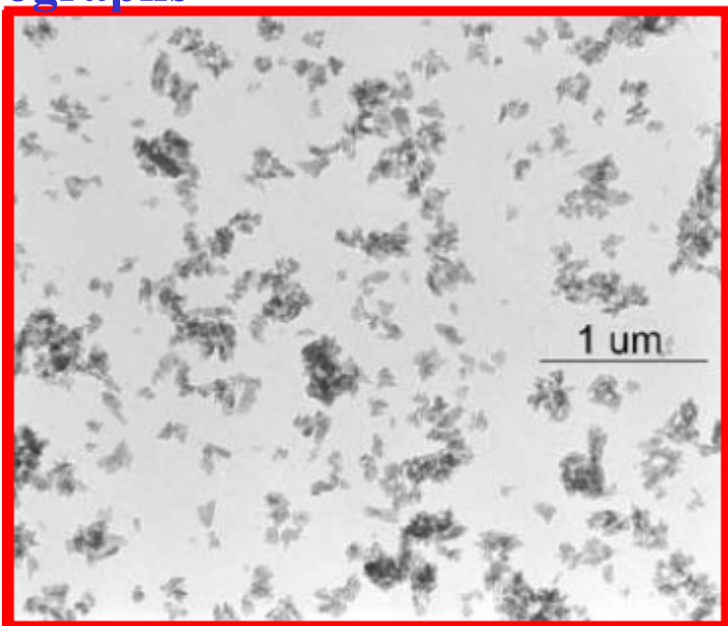
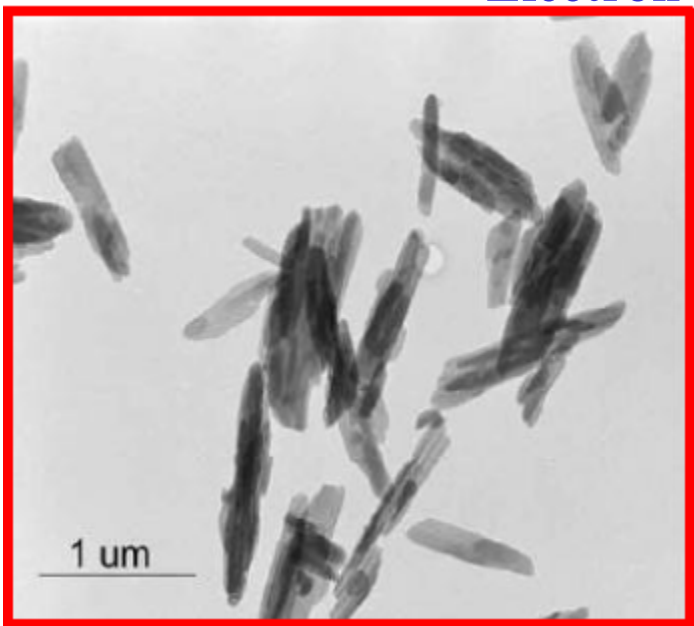
β form – bluish violet

Pigment Crystallography

Particle Size

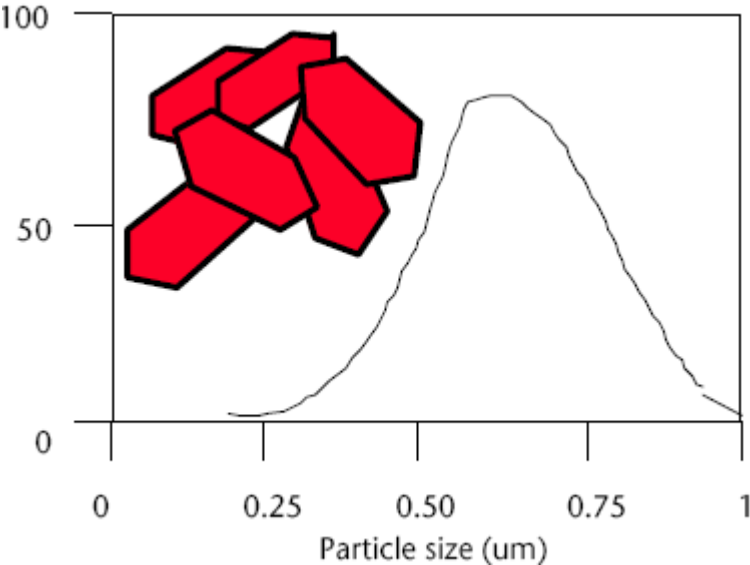
Crystal shape and size can be manipulated by both physical and chemical methods employed during the stages of manufacture and will play a key role in the end properties of the pigment.

Electron Micrographs

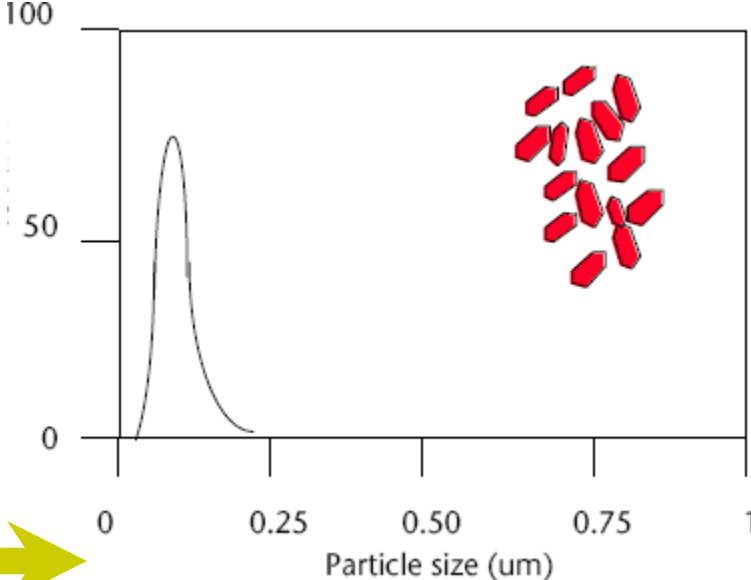


Pigment Crystallography

→ Δ□□P·MÄ ÖM



q ç □ M□□bIççM
≡ " ΔPçM
βΔ□□* M□□



f □·□□Ä □□M* .. □
' ·□□□
ç □Δ* □ Δ□M* P·
%*ç Ä ç □P□□çM

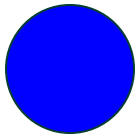
Pigment Crystallography

→Δ□ϑP·MÄϑMÄ* BÄ □·□ϑ□ϑP□

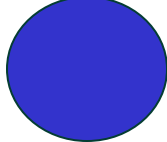
ø ϑ · M□ f □·□□Ä □M* .. □ Ř□ق M□



Greenish



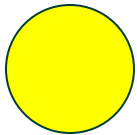
Cyan



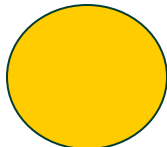
Reddish



Greenish



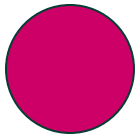
Yellow



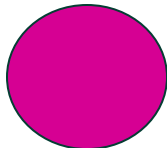
Reddish



Yellowish



Magenta



Bluish

ć □Λ* □Λ M* □ΛΛΛΛΛ □Λ* □Λ M* PΛ ΛΛΛΛΛΛΛΛ ≡ ” Λ□β M

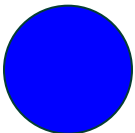
Pigment Crystallography

→Δ□□P·MÄ 0MÄ* BÄ·~ □PΔ·Ä□□ M□□M□

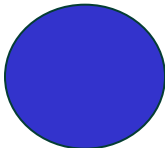
ø ȳ · M□ ☺ M□·□_ ~ R□ق M□



Greenish



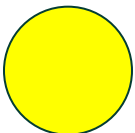
Cyan



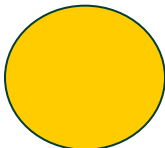
Reddish



Greenish



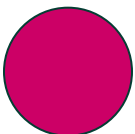
Yellow



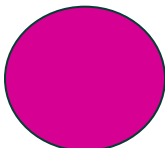
Reddish



Yellowish



Magenta



Bluish

R□ق M□ q ÷ □ΔH□□ ø ȳ · M□

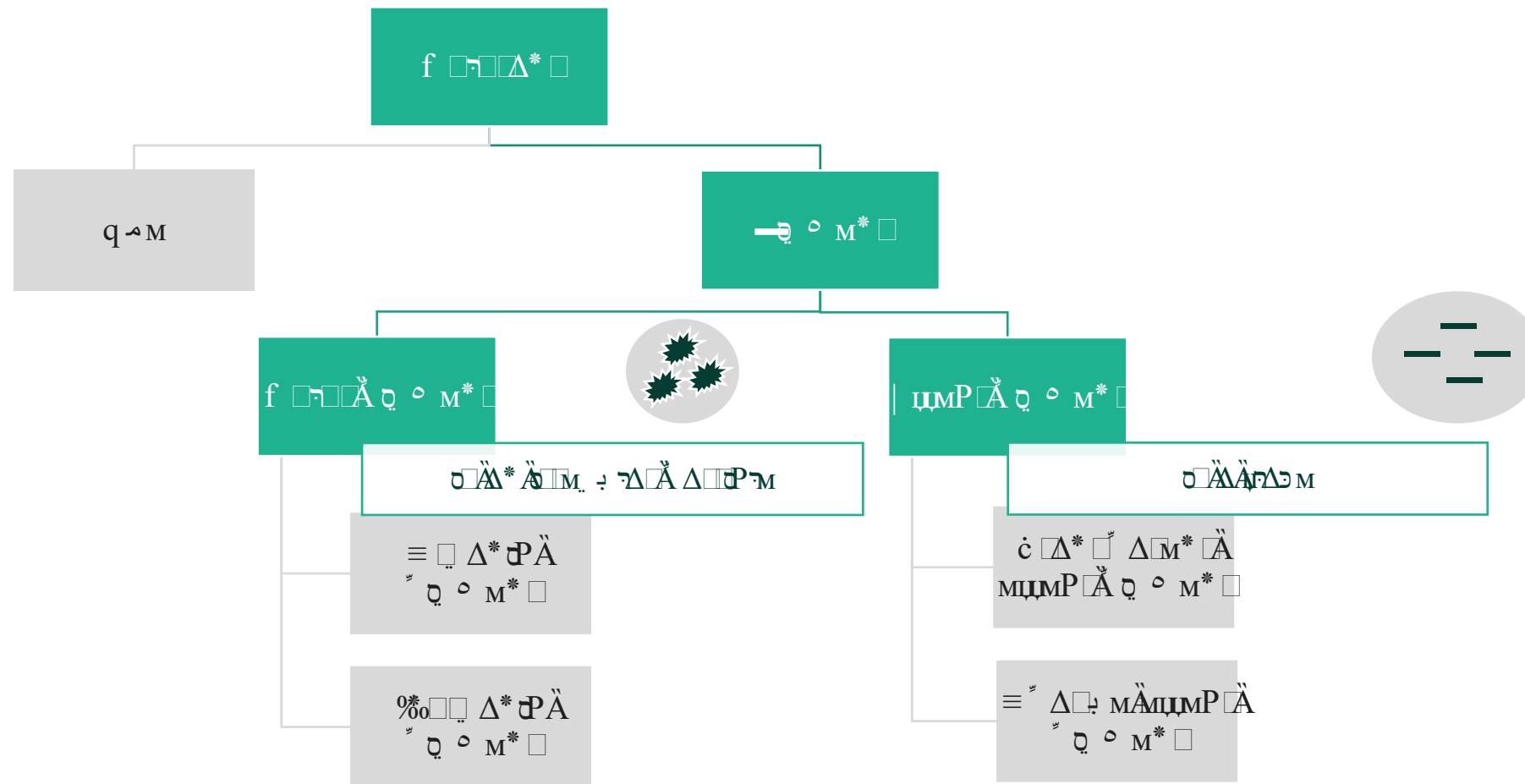
Effect Pigments



What is the difference between color and effect pigments?

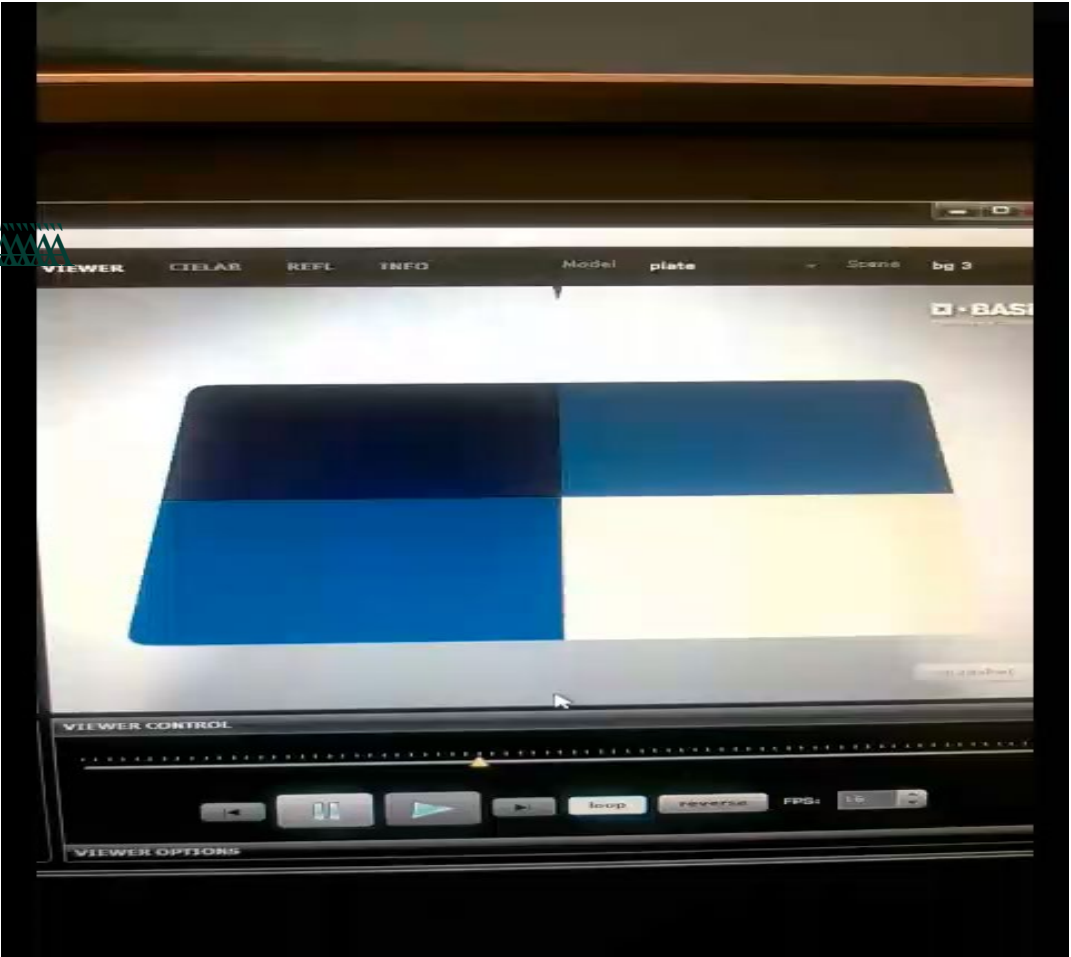


What is the difference between color and effect pigments?



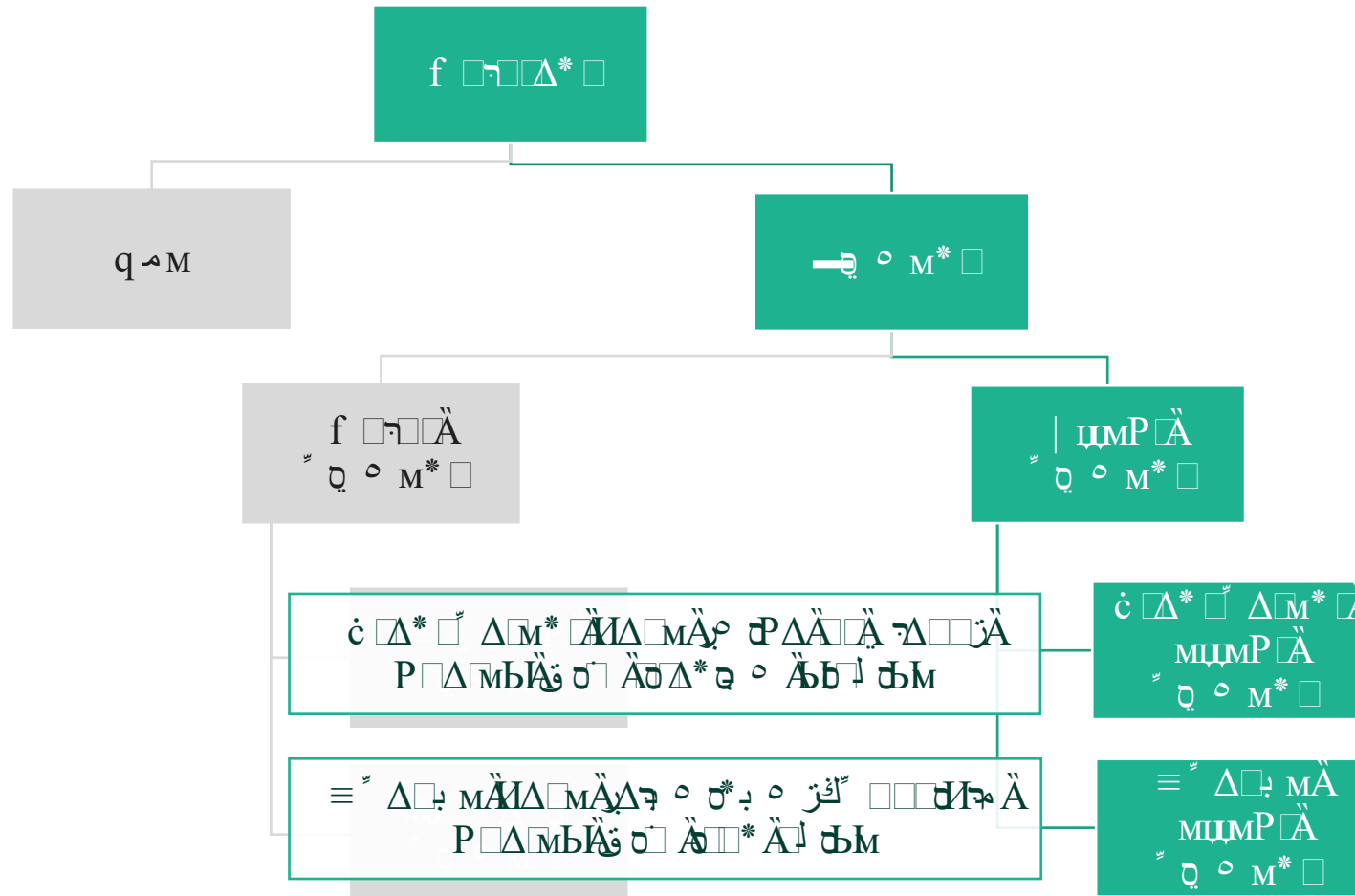
What is the difference between color and effect pigments?

ø M□. M*ñÄ
^ ǎ MÄÄ' Ä ÄÄÄÄ
پس لکے آہیں

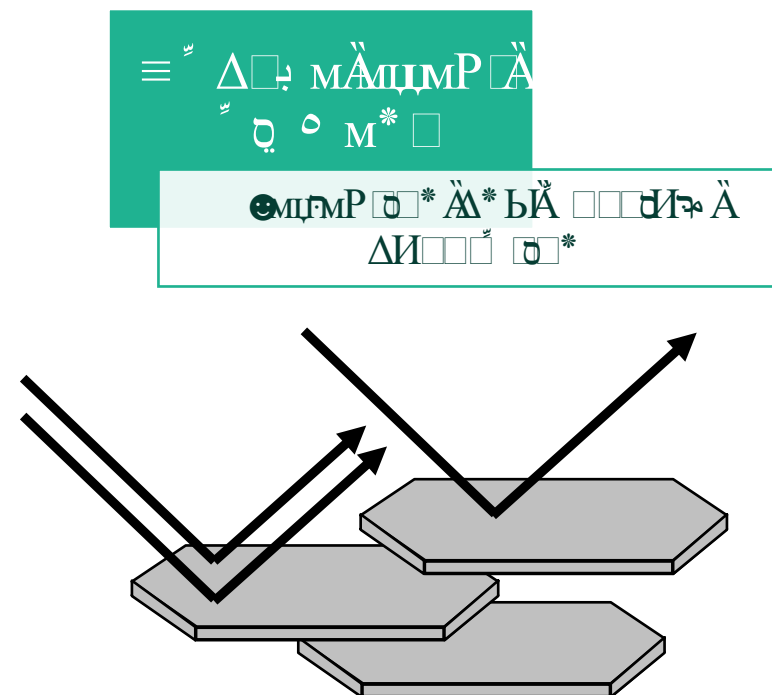
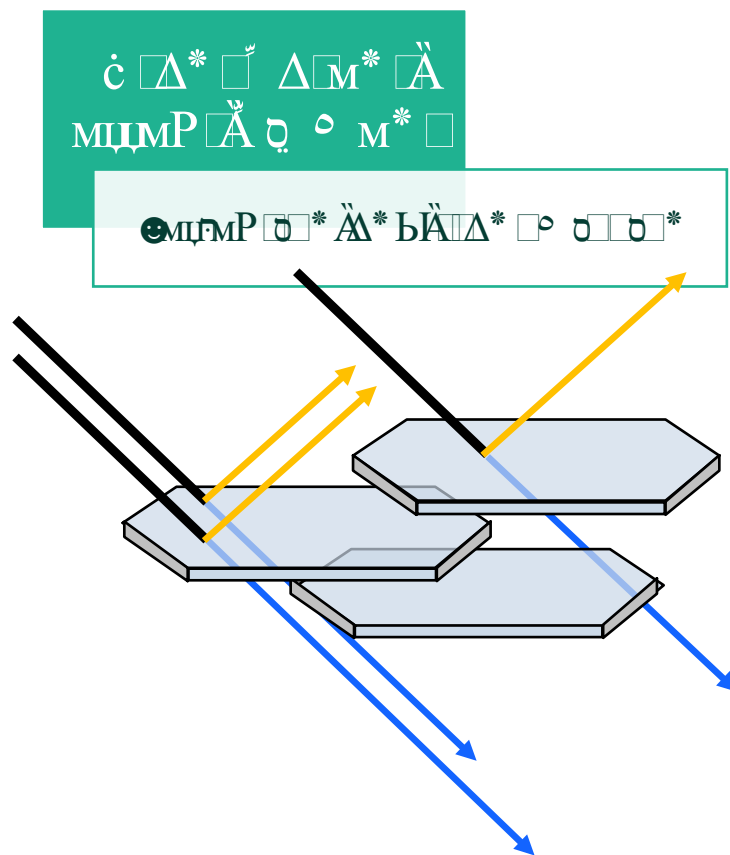


Řé ° σ ΔñÄ
☿ □ΔÄ' □Ä
^ ǎ MÄÄ' ø Ä
پس □Δ□MÄÄ
° PΔž

What is the structure of effect pigments?

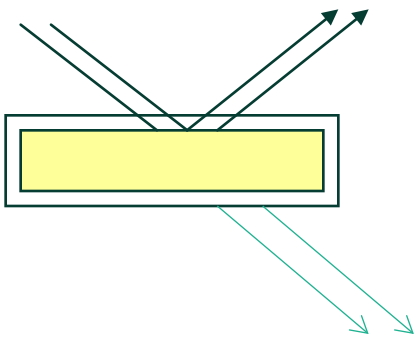


How do effect pigments function?



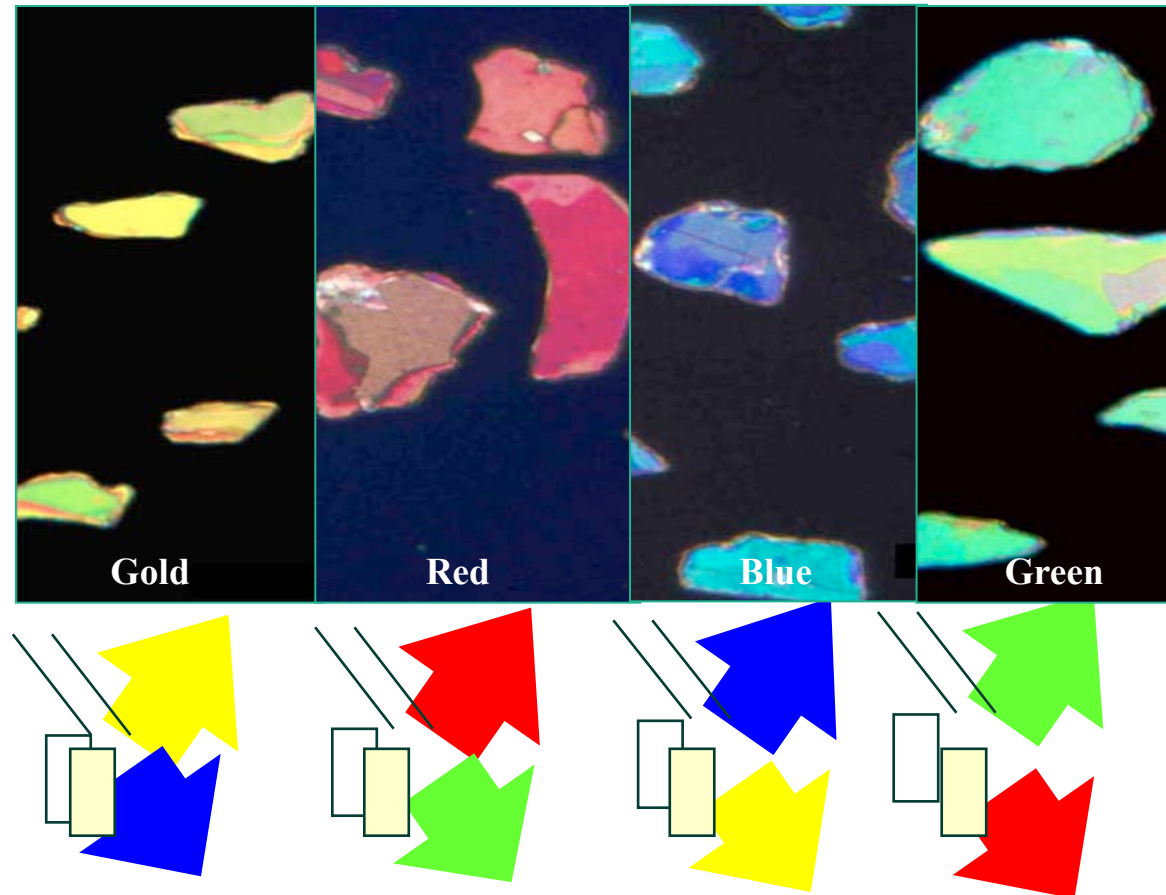
Transparent effect pigments e.g. based on mica

- βϒϒMÄ□□ ÄϒM□Ä □* Bβ*..
- ç □Δ* □ ΔM* □Ä□□ ÄM□ΔP □MÄ* BM
- %□□□MÄ* ÄÄ" " ϒΔ□□* Ä mBb □
- Bβb□□* ΔϒP□Δ□□* .. ÄMρM□□Δ□
➤ M□□□Ä □□* ÄM□□□□Ä □□□

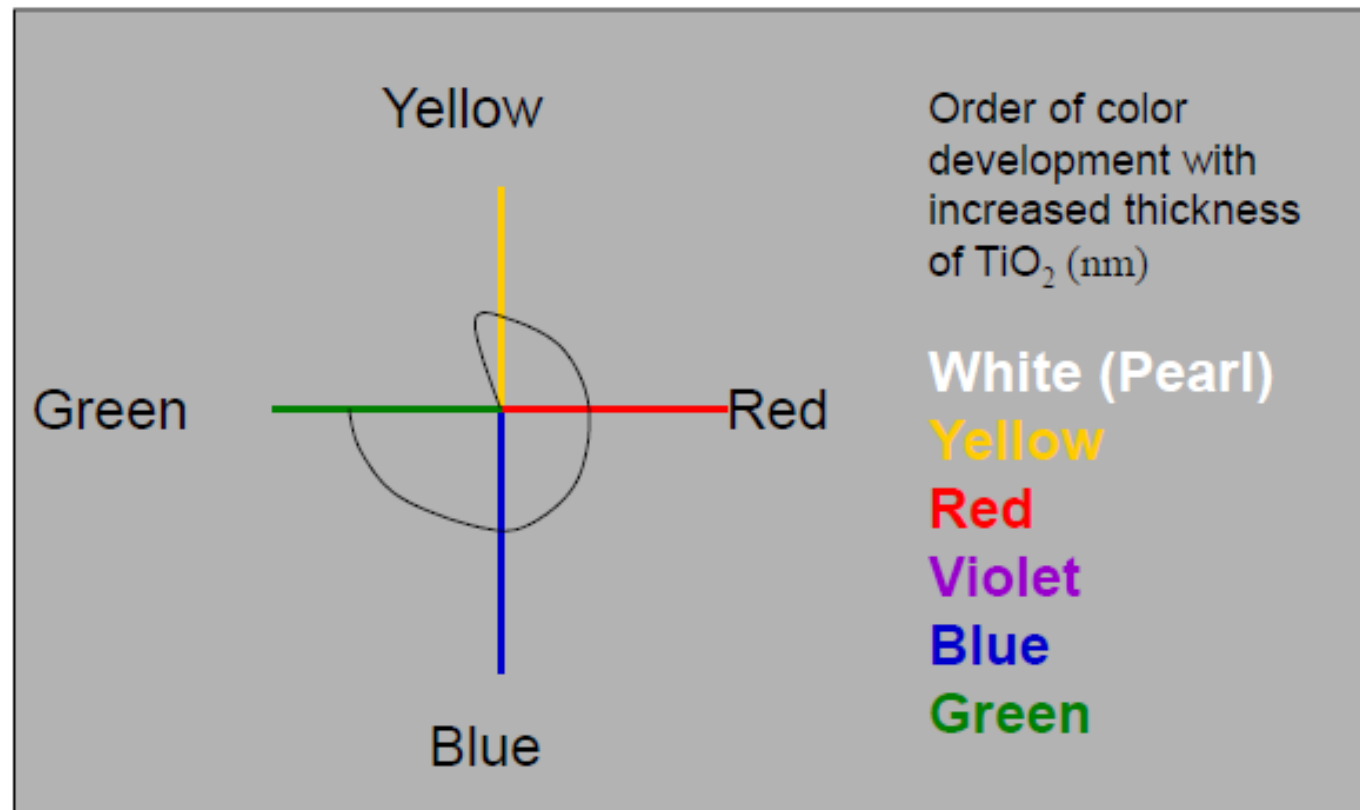


Transparent effect pigments

Influence of coating thickness

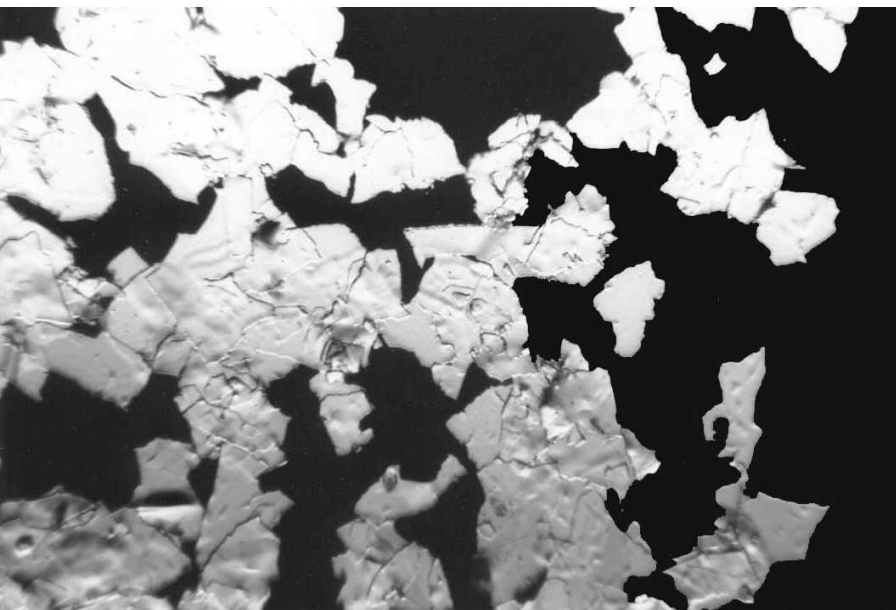


Thickness of layer determines the interference color



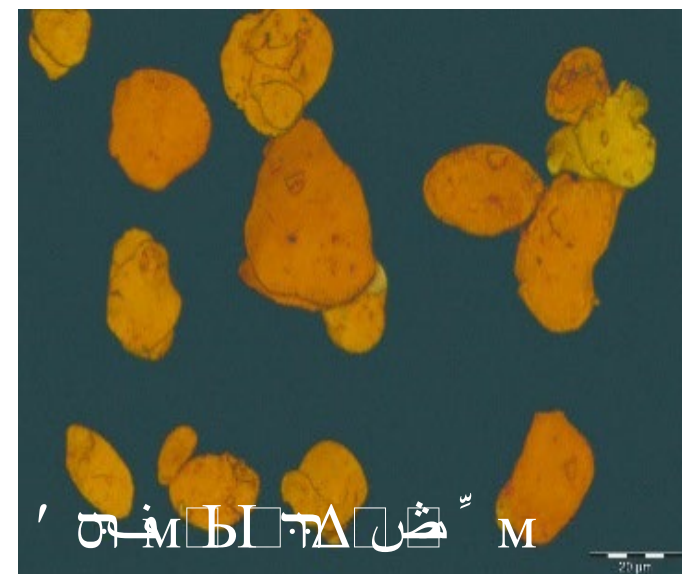
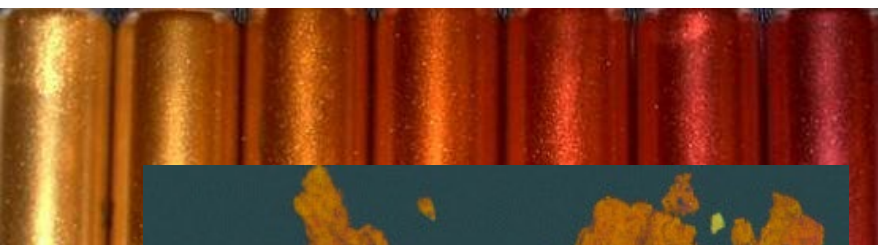
Opaque effect pigments – aluminum

- [illegible]



Opaque effect pigments – coated aluminum

- [illegible]



Effect Pigment Key Points

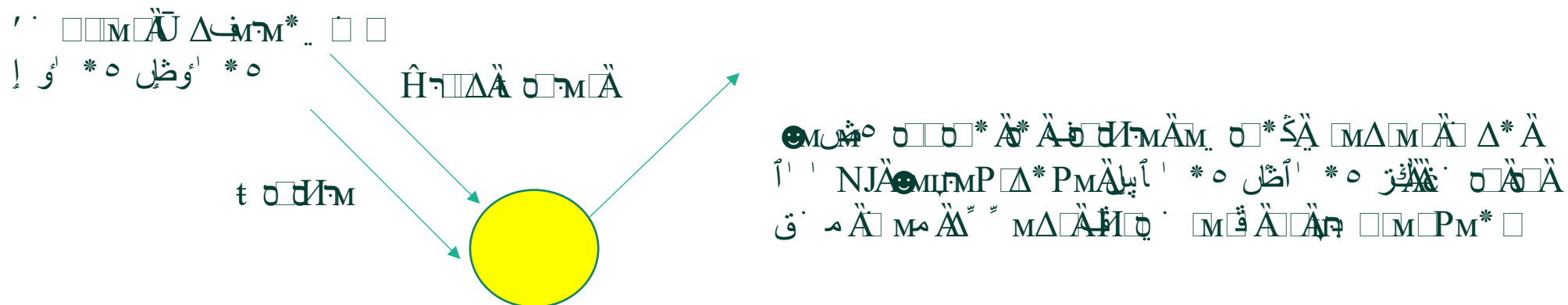
- ' □Ä*Ä* ʔ ʔ
 - ø ʔ ' Ä □ MΔ□Ä ʔ ʔ*.. Ä ʔ □ Ä MβΔÄ□Ä Ä□□Ä ʔ ʔ□Ä ʔ ʔ□Ä M□□□□Ä □ MÄMMP ʔ
 - ' M* M□Δ□Ä□Ä□ M ʔ Ä Δ□□P□MÄ□MÄ□M□Δ□□MÄ□Ä□Δ* ʔΔ□Ä* ʔ □ ʔ



Fluorescent Pigments



Visual Model

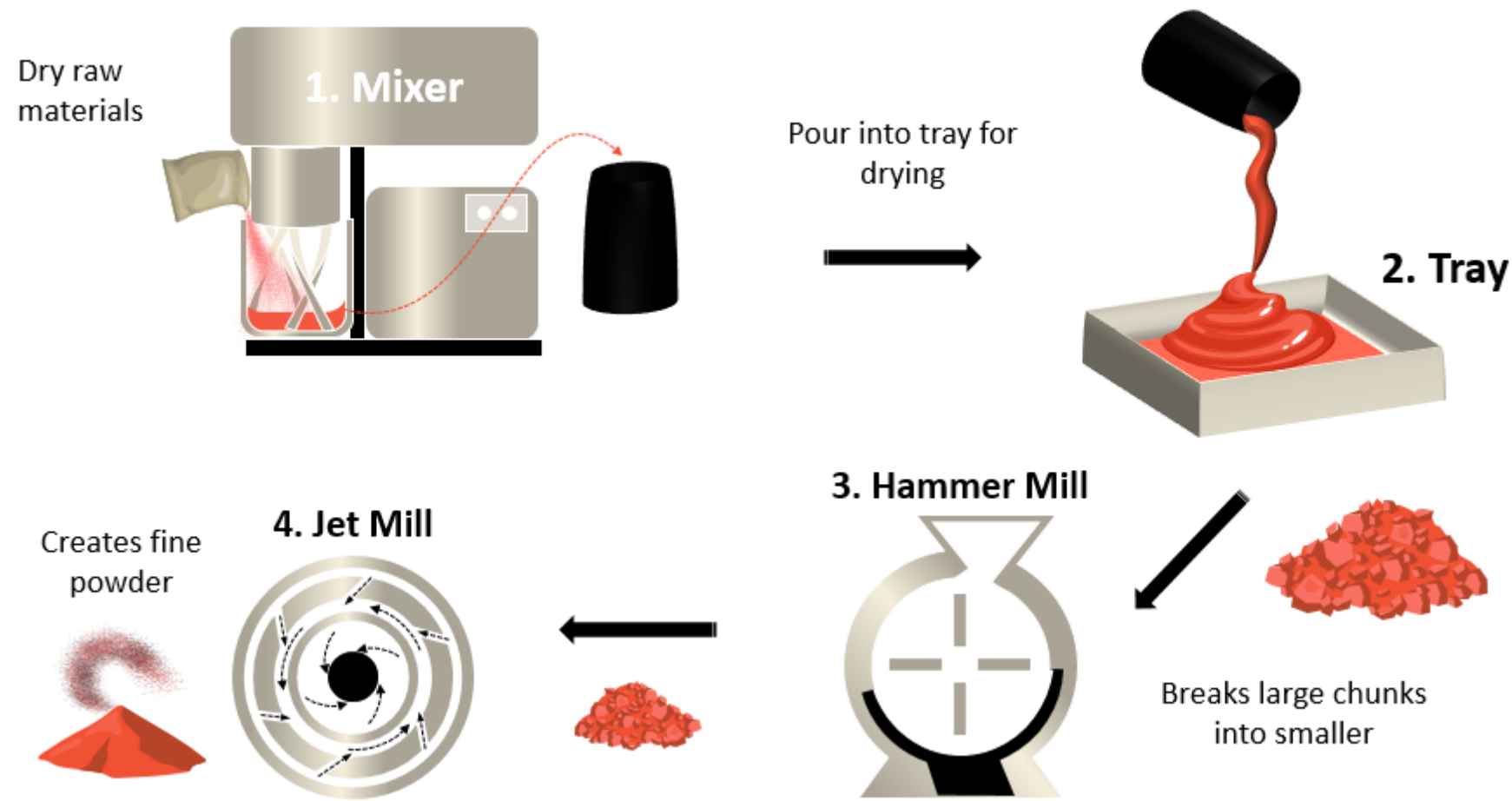


Fluorescent Pigment Manufacturing Technology

- ^ þ ƒ Ä—ƒ P□* BM* □Δ□*
- ' þ □ M* □□* Ä—ƒ □ M□□Δ□*
- | □ □□* Ä—ƒ □ M□□Δ□*
- f □□□Δ□* Ä PPþ □□Ä þ □*.. Ä—ƒ □ M□□Δ□*
- Θ□□Ä □M□Ä—ƒ □ M□□Ä□□ □ mBI



Production



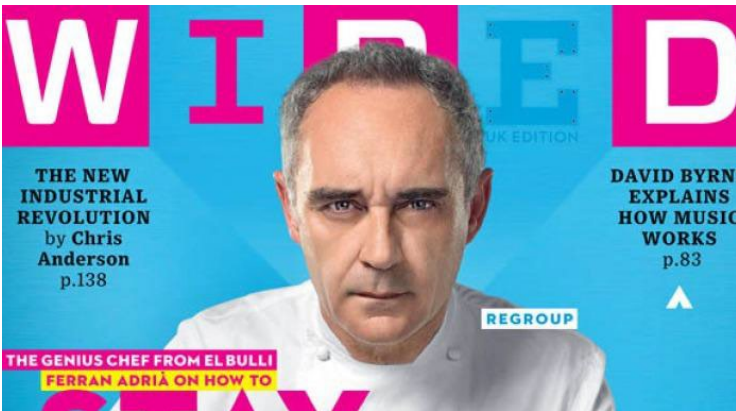
Fluorescent Pigments-Applications

- —Aσ* □□
- —□□ B M □ A □ Δ σ* .. □
- %ω □
- —Δ □ □ P □
- ċ M l □ M □
- ≡ □ M □



Fluorescent Pigments For Inks

- R0 □ □ Δ' · PÄ00 □
 - f □*fM* □* ΔÄ* bÄHt
- ß M □Ä00 □
 - Ü ΔM□Ä □fM* □Ä* bÄHt
- ' □Δf □MÄ00 □
 - Ü ΔM□Ä □fM* □
- q 0 □ ΔÄ00 □
 - ' □fM* □



Fluorescent Pigments: Many Chemistries + Applications

- $\wedge$ □□ΔbÄ* .. mÄüÄ · m° 0□□0m□éMP * □·□.. 0m□
- ü Δ*~Ä " " TPΔ0□* □
- %Ä%Ä% " □□Δ* Ä □Ä mMPÄ · mÄ · Ä—□bP Ä □Ä
ć · mÄ " " TPΔ0□*
- ≡ * mÄ 0mÄ □m□Ä □□Ä0Ä 7
- ' 0□Ä* 7

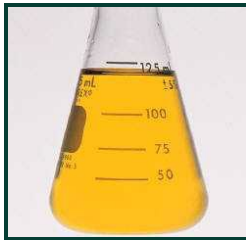
How are colorants used in inks?



How do you make sure the pigment sticks to the substrate?



→ ◊ M* □



q ◊ m□◊*.. Ä. M* □Ä
◊□◊* Ä bbb◊m

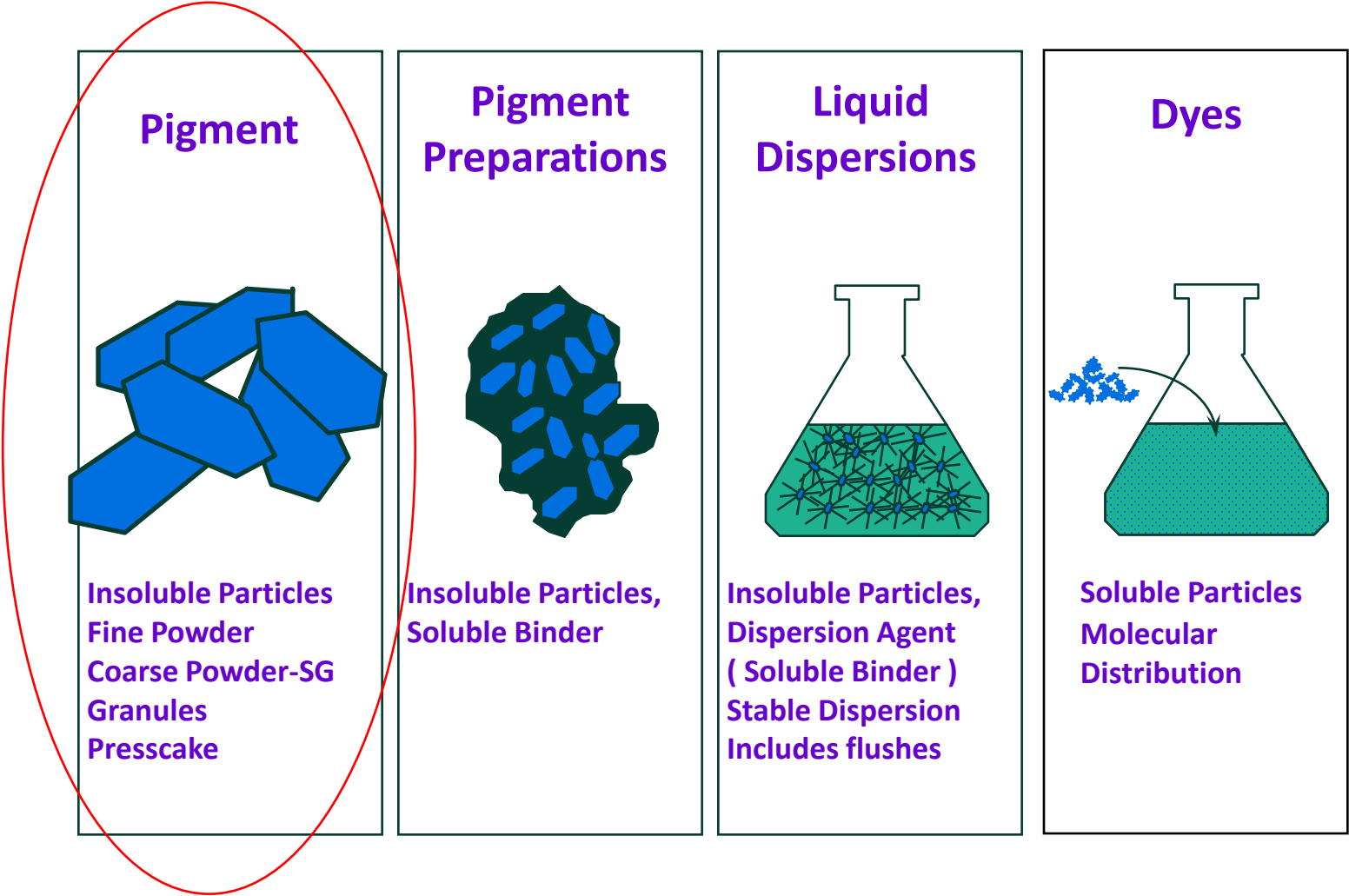


q ◊ m□◊* Ä □Pm□□Ä
M□Ä bbb□□m

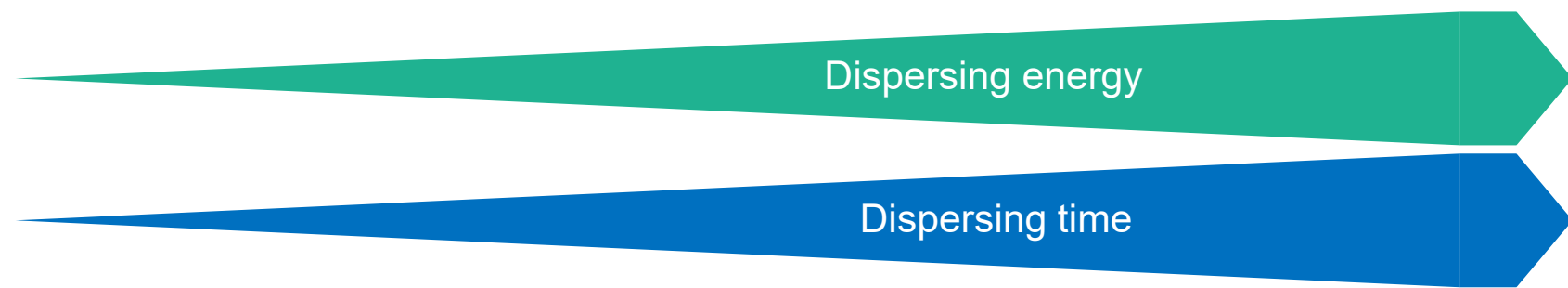


→ ◊ M* □Ä Δ□M□
f □Δ◊*.. ÄÄ◊

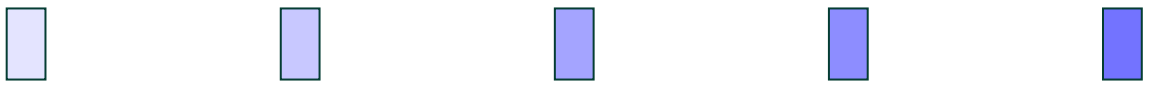
Product Forms



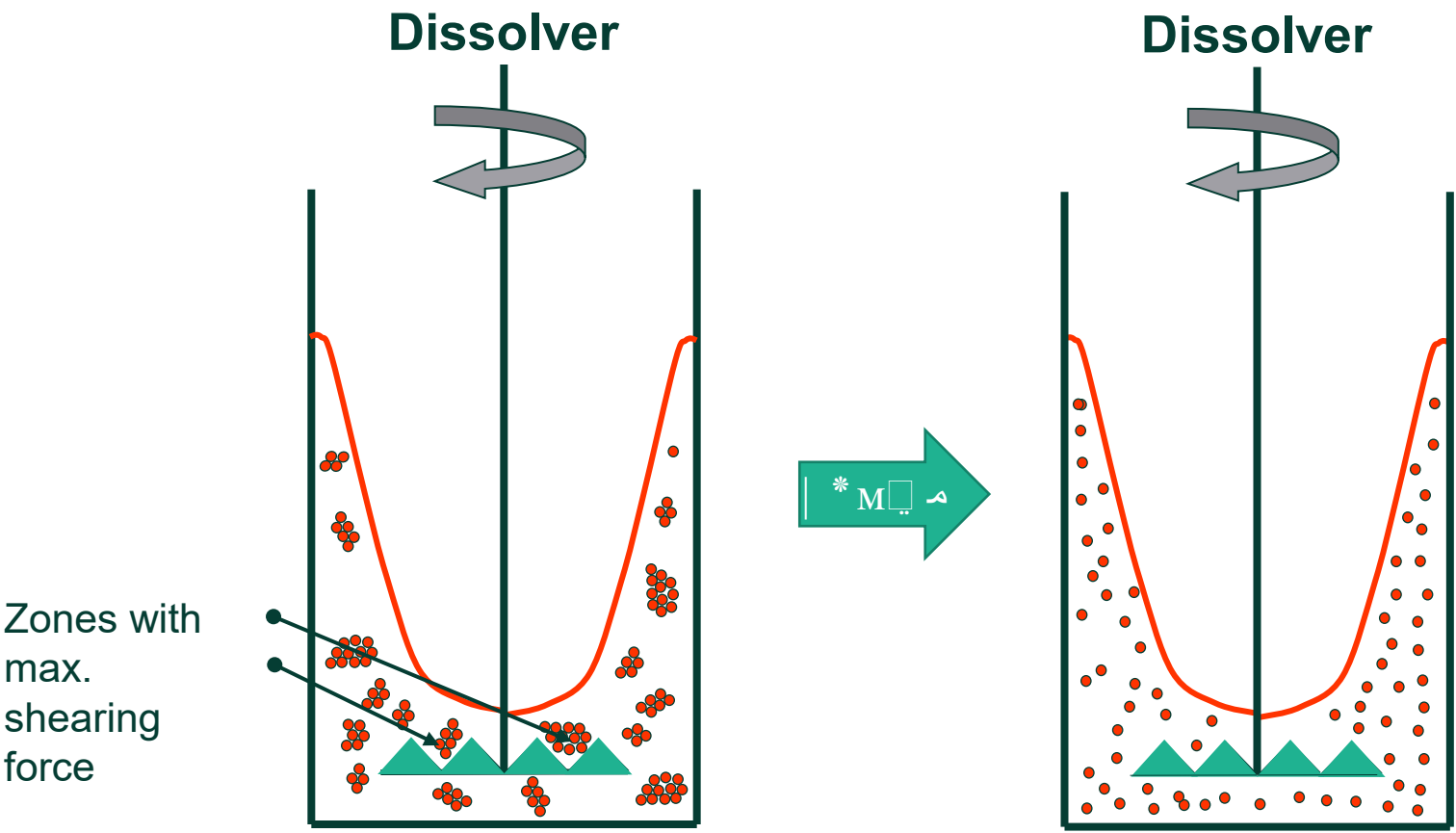
Dispersion process



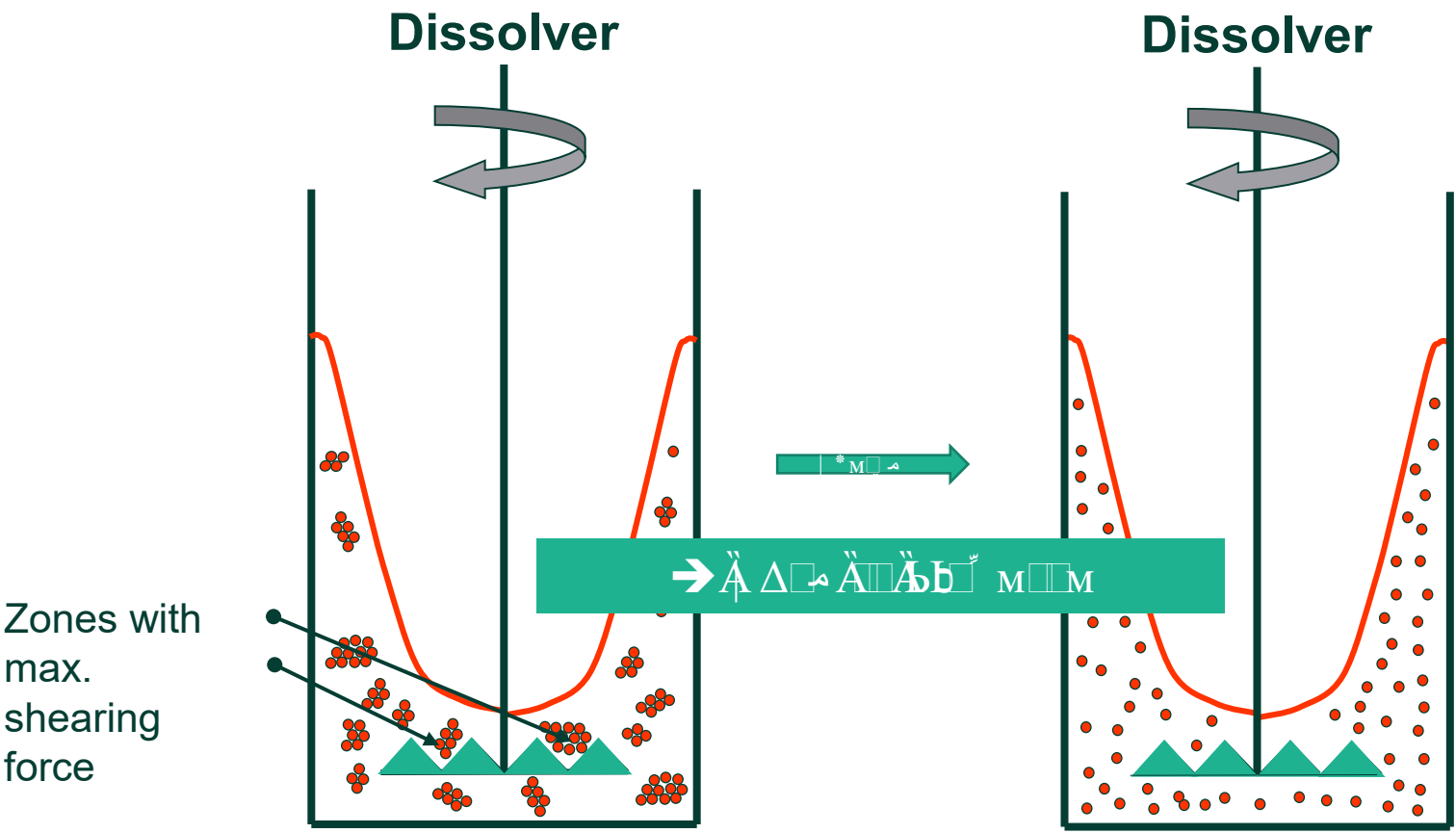
White reductions of the dispersed blue pigment at different stages of the dispersing process:



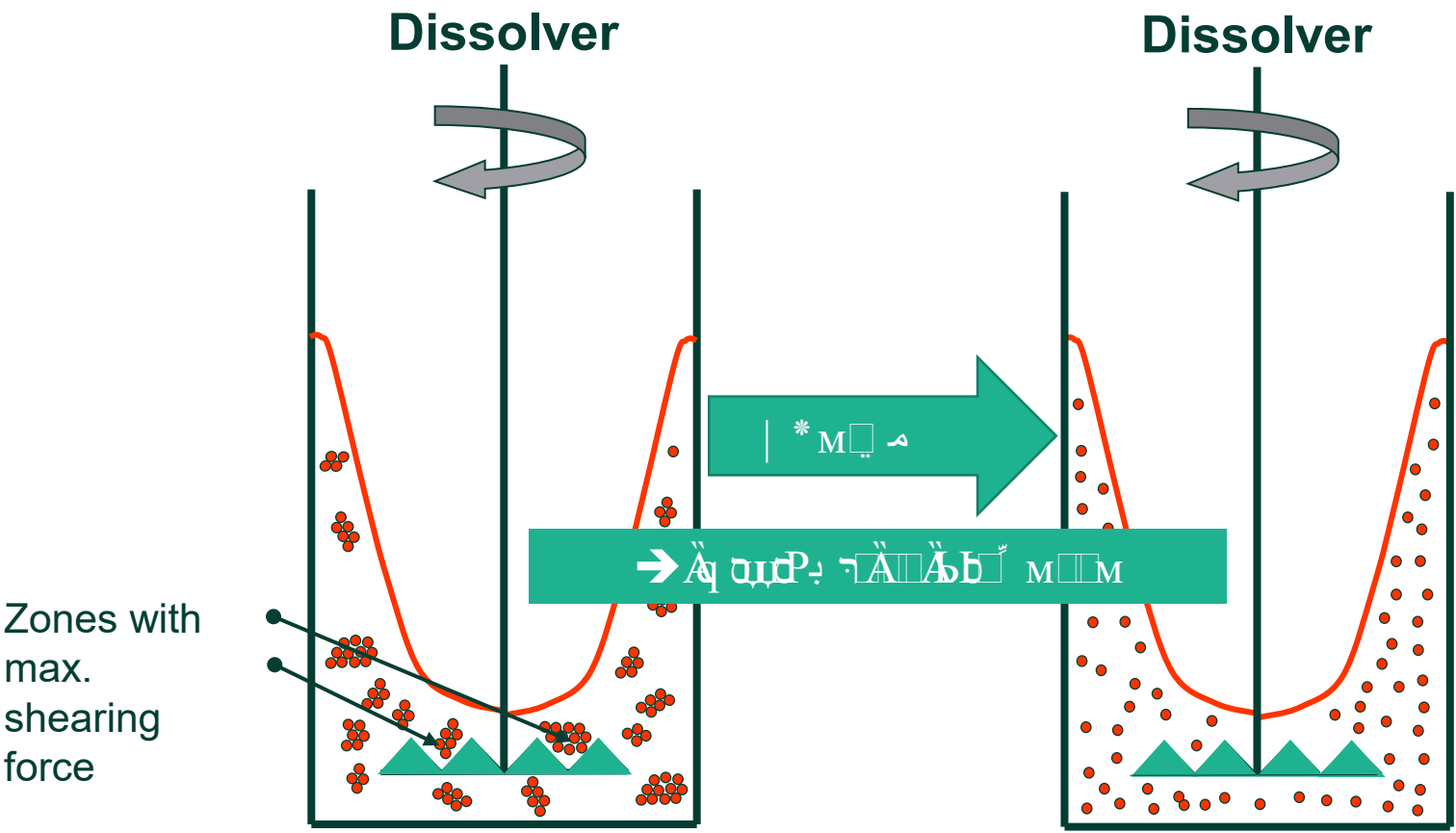
Dispersing with a dissolver



Dispersing with a dissolver



Dispersing with a dissolver



Types of Dispersing Equipment

- $\bar{u}$ מַבְלֵאִי שָׁחַד
- $\emptyset$ אֶפְסוּר * אֲדַרְשֶׁה
- t מַתְּכֵם דַּרְשֶׁה
- $\wedge$ אֶמְצֵא מַעַל
- $\bar{i}$ אֶמְצֵא מַעַל שָׁחַד
- f אֶמְצֵא מַעַל * אֶמְצֵא מַעַל * אֶמְצֵא מַעַל * אֶמְצֵא מַעַל *

חַסְדֵּי אֱלֹהִים מְבָרָכִים לְעוֹלָם וָעוֹלָם

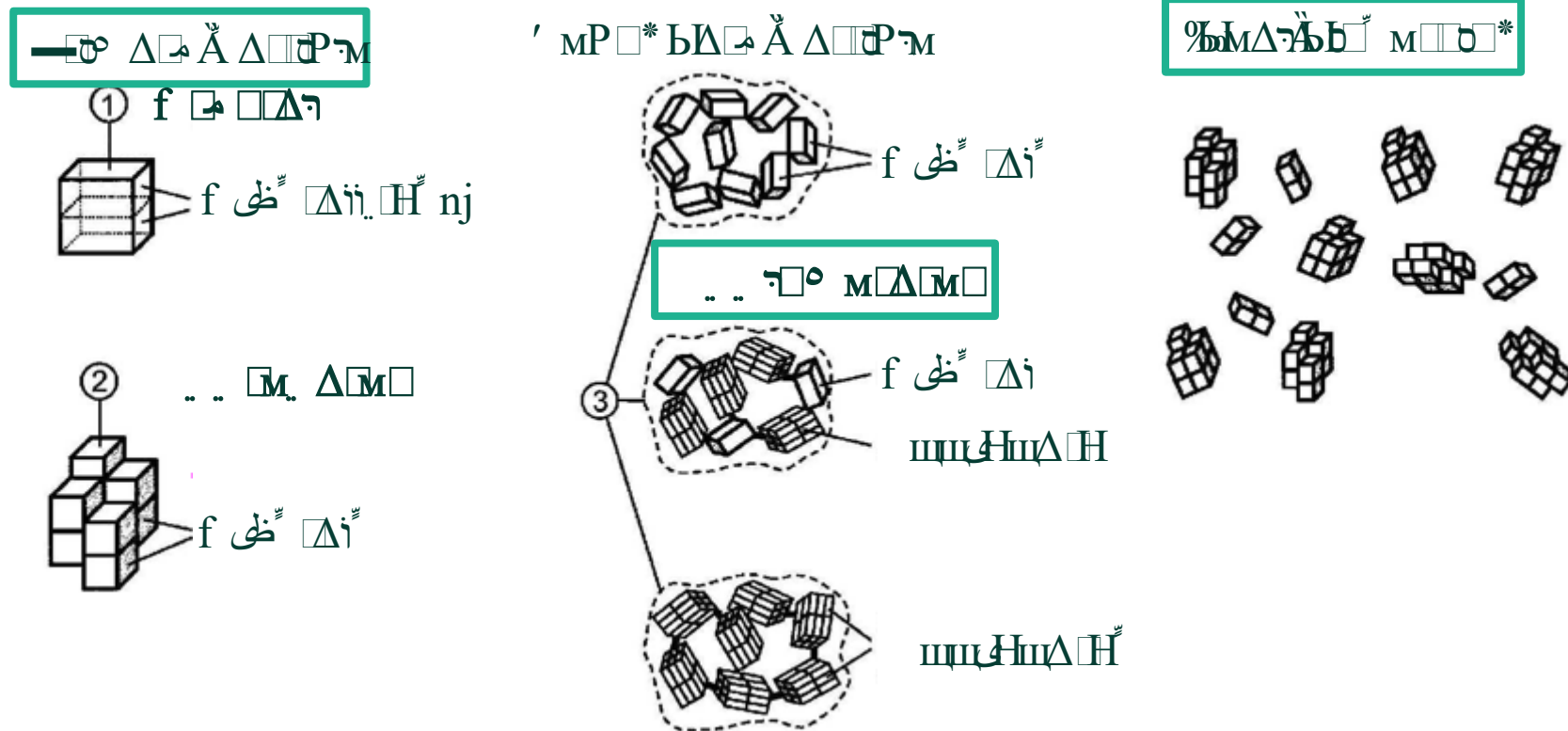


Courtesy of Buhler



אֶת־אֲנִי
וְהִסֵּם

The pigment model (DIN 53206)



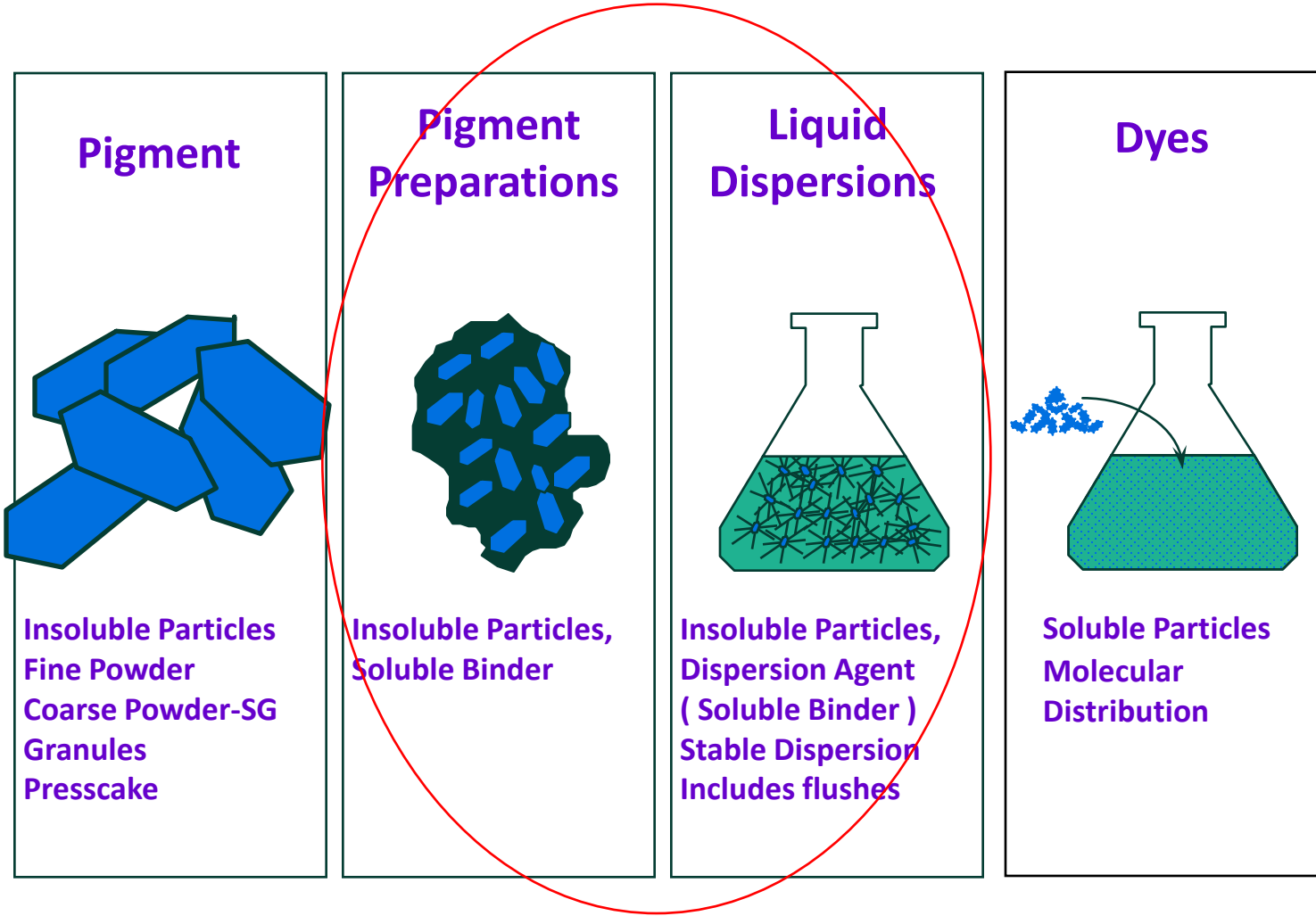
How do properties of organic and inorganic pigments relate to particle size?

Property	Organic 	Inorganic 
Particle size		
Transparency		
Opacity		
Color strength		
Chroma		
Fastness properties		
Dispersibility		

Pigment Preparations

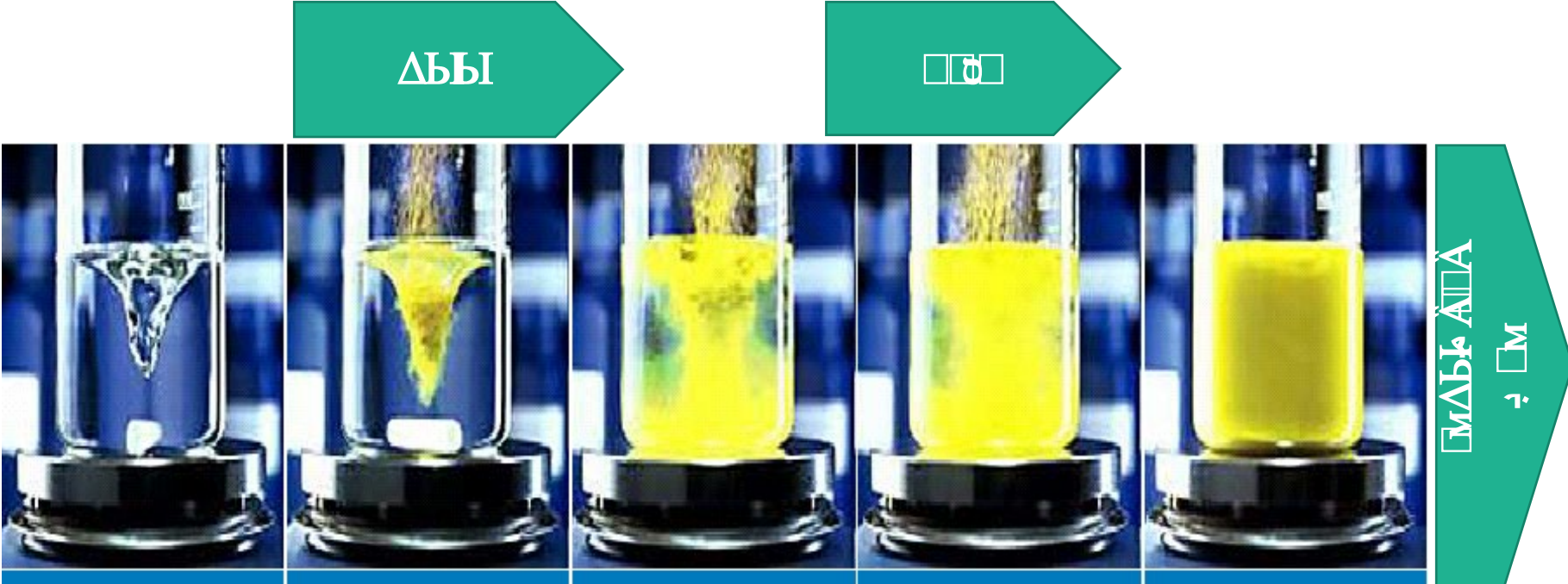


Product Forms



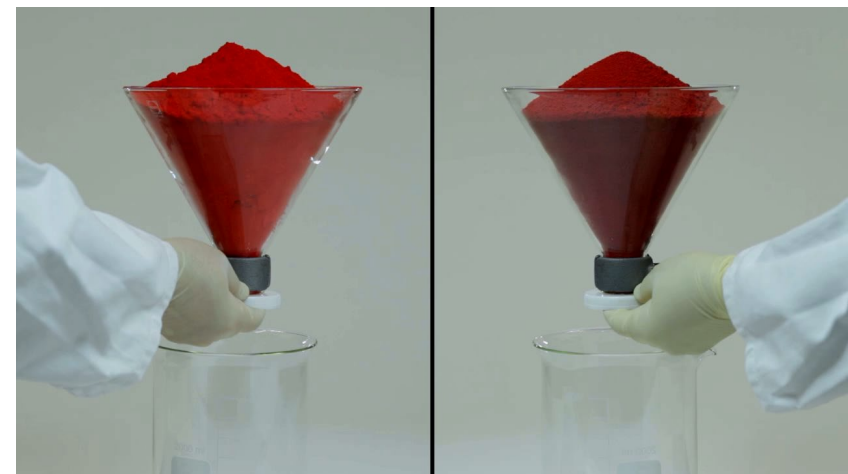
What are pigment preparations?

- — ċ M* Ä M' ΔΔÖ* ÄMÄ MÖB' MMBÄ ċ M* Ä* ÄTÄÄÖB' TÄMÖMÄÄ ċ
- Θ□□ ΔTÄÄ ċ M* Ä M' ΔΔÖ* ÄMÄ □* □ض ċ M* MBI



Properties of preparations

- $\Theta \square^* \text{بلى} \square \square^*$..
- $| \Delta \square \text{م} \ddot{A} \square \ddot{A} \text{بلى} \square \square \text{م}$
- $| \Delta \square \text{م} \ddot{A}^* \text{بلى} \ddot{A} \text{بلى} \text{P} \square \ddot{A}^* \text{P} \square \square \square \square \Delta \square \square^*$
- $\beta \Delta \square \ddot{A} \text{P} \text{م} \text{م}^{\circ} \text{م}^* \ddot{A} \text{P} \square \text{بلى} \ddot{A} \square \square \text{م}^* \text{..} \square \ddot{A}^* \text{بلى} \text{P} \square \square^{\circ} \Delta$

$$q \in \mathbb{A} \circ M^* \quad \text{---} M^* \Delta \Delta \mathbb{D}^*$$


Benefits of Preparations

- ø ŵ´ ÄµµPµ*Pµ
- ø ŵ´ Äµ Δŵµ Ä* BÄM´ µµBµ Pµµµµ
- µµBµ Pµµ* ÄµÄB´ µµµµ* Äµµµ µµ
- ΘµÄµµµÄBµ µµ µ* ÄµPµµµΔµ
- | µPµµµ* Äµµµµµµµ* Äµµµ µµµ* ÄµÄÄ* BÄ µµBµ Pµµ*
- βΔµµÄµµµÄµ Δ* µµÄ µµµµµµ
- ΘµÄ µ* Bµµ µµÄ µ ŵ µ µ* Äµµµµµµ
- f µµÄµBµ Pµµ* Ä* Ä µµBµ Pµµ*



Key takeaways for ink formulators



What if you are asked to develop an ink?

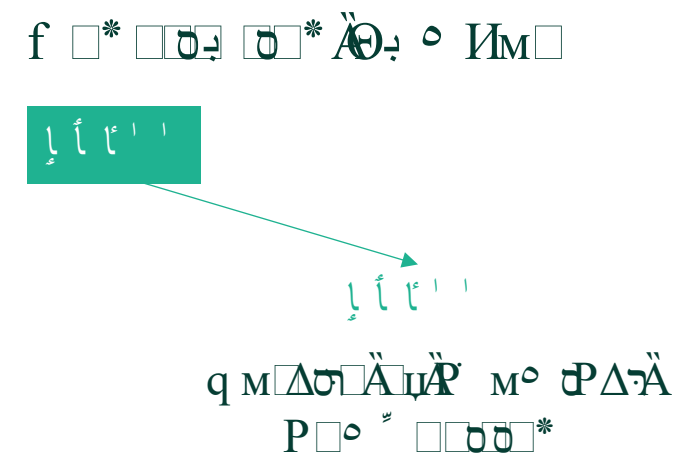
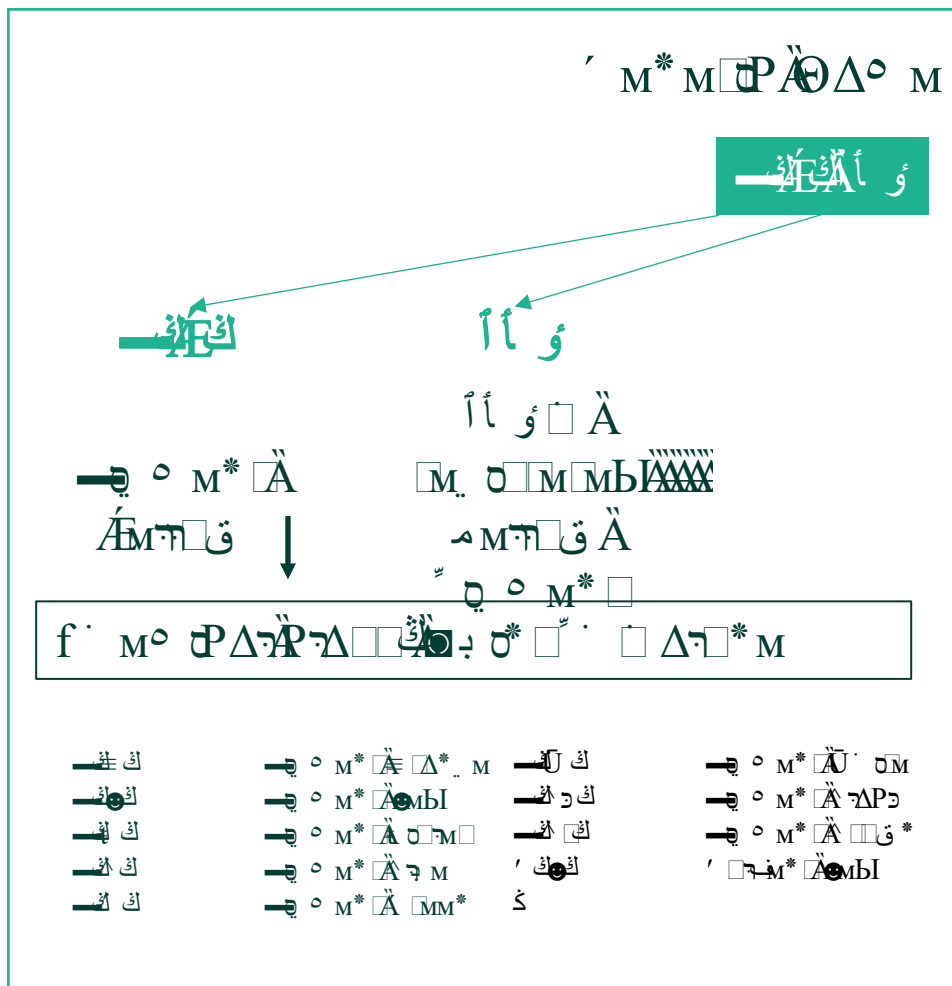
- [illegible]



Initial questions

- [illegible]

What is Color Index?



Initial questions

- Ū · ΔÄ □□Ä*BM
- f □□Ä ΔPM
- Ū MΔ□ MÄ* BÄ · □Δ□* M□ÄM□ ◊ M° M* □□
- ☹M. ◊ Δ□□ÄM□ ◊ M° M* □□
 - — Ä◊ Δ□Ä □□° ΔPÄ ° σ* M□
 - ø MΔفÄ MΔ□
- Ū · ΔÄ' MÄ□Ä* ◊Ä' □M°
- Ū · ΔÄ' MÄ□Ä □* □* .. Ä □□PM□□

C.I.Pigment Violet 19



Cinquasia
Red
L4100 HD

Cinquasia
Violet
L5120

Mass Tone

Printing Overview

	Gravure	Flexo	Lithography	Screen	Digital
Typical Print Speeds	Up to 600 m/min	~ 450 m/min	600-700 (heatset) 15000-18000 iph (SF)	100 m/min (rotary)	30-100 m/min (state of art)
Typical Ink Viscosities	20-25 secs Zahn 2 0.05 – 0.2 Pa.s	20-25 secs Zahn 2 0.05 – 0.5 Pa.s	40-100 Pa.s	1 – 5 Pa.s (flatbed) Lower for rotary	6-15 mPa.s solvent/UV 2-5 mPa.s aqueous
Typical Wet Film Thickness (μm)	7 – 15	3 – 7	0.5 – 2	8 – 30	6 – 12 (multiple passes)
Typical Particle Size	< 8 μm	< 4 μm	< 4 μm	-	Mean of 0.05 – 0.2 μm Max. of 0.5 – 1.0 μm
Main Binders	Nitrocellulose, Cellulose Acetate Propionate, Polyvinyl butyral, Polyamide, Polyurethane, Polyvinyl, Acrylics , Acrylated Oligomers	Nitrocellulose, Cellulose Acetate Propionate, Polyvinyl butyral, Polyamide, Polyurethane, polyvinyl, Acrylics , Acrylated Oligomers	Modified rosin esters (phenolic), Hydrocarbons, Drying oils (linseed oil, alkyds) Acylated Oligomers	Polyvinyls, Plastisol, Acrylated Oligomers	Polyvinyl, nitrocellulose , Acrylated Oligomers
Main Solvents	Alcohols (ethanol, n- propanol), Esters (ethyl acetate, n-propyl acetate), glycol retarders (ethoxy propanol), Water Ketones , Acrylated Monomers	Alcohols (ethanol, n- propanol), Esters (ethyl acetate, n-propyl acetate), glycol retarders (ethoxy propanol), Water Ketones , Acrylated Monomers	Petroleum distillates (heatset inks, 260-290°C) , Mineral oils, Vegetable Oils (linsed, soya, rapeseed etc) Acrylated Monomers	Glycols , Acrylated Monomers	Water , Glycols , esters , ketones , Acrylated Monomers

Initial questions

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From the Pigment Technical Data Sheet...

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Final Thoughts

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